

## Section Review Answers

### Chapter 2

#### Section 1

1. Matter has mass and occupies space. Light has neither mass nor volume, nor is it made up of atoms.

2. An atom is the smallest unit of an element that has that element's properties. Both atoms and elements are matter.

3. A molecule is the smallest unit of a substance that keeps all of the physical and chemical properties of that substance. Oxygen ( $O_2$ ) is an example of a molecule formed by atoms of one element. Water ( $H_2O$ ) is an example of a molecule formed by atoms of two elements.

4.  $H_2O$

5. elements, compounds

6. a. element  
b. compound  
c. compound  
d. element

7. Both can contain atoms of more than one element. A pure substance, such as water, has the same composition throughout and definite properties. The composition of a mixture, such as air, can vary because a mixture contains more than one pure substance.

8. David's statement reflects the common meaning of "pure." Susan's use of the word "pure" is more correct scientifically because honey is a mixture

# Chapter Section Review Answers

**Jeffrey S. Nevid, Spencer A. Rathus**



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