



# CHEMISTRY

## CHEAT SHEET



# NAMING SIMPLE COMPOUNDS

## NAMING SIMPLE COMPOUNDS CHEAT SHEET

**Naming Compounds with Their Formulas**  
(Show the ions involved in the compound)



**Directions for naming:**

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is a molecular compound, use prefixes to indicate the number of atoms of each element.
6. If the compound is an acid, use the prefix H- followed by the name of the anion.

### Binary Compounds

First Element	Anion
H	hydrogen
C	carbon
N	nitrogen
O	oxygen
F	fluorine
Cl	chlorine
S	sulfur
Br	bromine
I	iodine
Al	aluminum
Fe	iron
Cu	copper
Zn	zinc
Ag	silver
Au	gold

### Common Ions and Their Names

Formula	Name
Na <sup>+</sup>	Sodium
K <sup>+</sup>	Potassium
Ca <sup>2+</sup>	Calcium
Mg <sup>2+</sup>	Magnesium
Al <sup>3+</sup>	Aluminum
Fe <sup>2+</sup>	Iron (II)
Fe <sup>3+</sup>	Iron (III)
Cu <sup>2+</sup>	Copper (II)
Cu <sup>+</sup>	Copper (I)
Zn <sup>2+</sup>	Zinc
Ag <sup>+</sup>	Silver
Au <sup>3+</sup>	Gold (III)
Au <sup>+</sup>	Gold (I)
Cl <sup>-</sup>	Chloride
Br <sup>-</sup>	Bromide
I <sup>-</sup>	Iodide
S <sup>2-</sup>	Sulfide
O <sup>2-</sup>	Oxide
N <sup>3-</sup>	Nitride
C <sup>4-</sup>	Carbide

**Naming Compounds with a Metal and a Nonmetal**



**Directions for naming:**

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.

**Directions for naming:**

Formula	Name
NaCl	Sodium chloride
KCl	Potassium chloride
CaCl <sub>2</sub>	Calcium chloride
MgCl <sub>2</sub>	Magnesium chloride
AlCl <sub>3</sub>	Aluminum chloride
FeCl <sub>2</sub>	Iron (II) chloride
FeCl <sub>3</sub>	Iron (III) chloride
CuCl <sub>2</sub>	Copper (II) chloride
CuCl	Copper (I) chloride
ZnCl <sub>2</sub>	Zinc chloride
AgCl	Silver chloride
AuCl <sub>3</sub>	Gold (III) chloride
AuCl	Gold (I) chloride

**Common Names for Specific Compounds**

Formula	Common Name
NaCl	Table salt
KCl	Saltpetre
CaCl <sub>2</sub>	Plaster of Paris
FeCl <sub>3</sub>	Iron rust
NaOH	Sodium hydroxide
KOH	Potassium hydroxide

# Cheat Sheet For Naming Compounds

**Chemical abstracts**



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Chemistry Workbook For Dummies Peter J. Mikulecky, Chris Hren, 2014-11-26 Hundreds of practice problems to help you conquer chemistry Are you confounded by chemistry Subject by subject problem by problem Chemistry Workbook For Dummies lends a helping hand so you can make sense of this often intimidating subject Packed with hundreds of practice problems that cover the gamut of everything you ll encounter in your introductory chemistry course this hands on guide will have you working your way through basic chemistry in no time You can pick and choose the chapters and types of problems that challenge you the most or you can work from cover to cover With plenty of practice problems on everything from matter and molecules to moles and measurements Chemistry Workbook For Dummies has everything you need to score higher in chemistry Practice on hundreds of beginning to advanced chemistry problems Review key chemistry concepts Get complete answer explanations for all problems Focus on the exact topics of a typical introductory chemistry course If you re a chemistry student who gets lost halfway through a problem or worse yet doesn t know where to begin Chemistry Workbook For Dummies is packed with chemistry practice problems that will have you conquering chemistry in a flash *Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice)* Heather Hattori, Richard H. Langley, 2014-04-14 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time *U Can: Chemistry I For Dummies* John T. Moore, Chris Hren, Peter J. Mikulecky, 2015-07-21 Now you can score higher in chemistry Every high school requires a course in chemistry for graduation and many universities require the course for majors in medicine engineering biology and various other sciences U Can Chemistry I For Dummies offers all the how to content you need to enhance your classroom learning simplify complicated topics and deepen your understanding of often intimidating course material Plus you ll find easy to follow examples and hundreds of practice problems as well as access to 1 001 additional Chemistry I practice problems online As more and more students enroll in chemistry courses the need for a

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Webster's New International Dictionary of the English Language, Based on the International Dictionary 1890 and 1900 William Torrey Harris, Frederic Sturges Allen, 1911 *New International Dictionary*, 1920 An Introduction to Chemical Nomenclature R. S. Cahn, 2013-12-19 Introduction to Chemical Nomenclature R. S. Cahn, O. C. Dermer, 2013-10-22 Introduction to Chemical Nomenclature Fifth Edition delves into the nomenclature the system of how names or terms are formed of different compounds The book covers the development of chemical nomenclature the nomenclature of different ions salts and compounds under inorganic chemistry the principles involved in the nomenclature of organic compounds including hydrocarbons and heterocycles and special features and functional groups The selection also covers natural products such as carbohydrates lipids steroids amino acids and nucleic acids alkaloids and peptides as well as the miscellaneous chemical nomenclature which includes organometallic and isotopically modified compounds and polymers The text is a good reference for students who have trouble in the nomenclature of different chemical substances and those who want to study the principles behind the chemical nomenclature

*Inorganic Chemical Nomenclature* Lloyd McKinley, 1947 An Introduction to Chemical Nomenclature Robert Sidney Cahn, 1974 *Naming Chemical Compounds*, 2000 **Chemical Nomenclature** K. J. Thurlow, 1998 A general introduction to forms of chemical nomenclature dealing with systematic and trivial names Chapters are included on specialized naming systems for polymers and natural products and on the role of computers and the quest to find a quick and accurate naming program The Etymology of Chemical Names, 2019-10-08 Etymology of Chemical Names gives an overview of the development of the current chemical nomenclature tracing its sources and changing rules as chemistry progressed over the years This book is devoted to provide a coherent picture how the trivial and systematic names shall be used and how the current IUPAC rules help to reconcile the conflicting demands **The Naming and Indexing of Chemical Compounds** Chemical Abstracts (Periodical), 1962 **The Naming and Indexing of Chemical Compounds** Chemical abstracts, 1962

**An Introduction to Chemical Nomenclature** Robert Sidney Cahn, 1964 **The 100 Most Important Chemical Compounds** Richard L. Myers, 2007-08-30 What is a chemical compound Compounds are substances that are two or more elements combined together chemically in a standard proportion by weight Compounds are all around us they include familiar things such as water and more esoteric substances such as triuranium octaoxide the most commonly occurring natural source for uranium This reference guide gives us a tour of 100 of the most important common unusual and intriguing

compounds known to science Each entry gives an extensive explanation of the composition molecular formula and chemical properties of the compound In addition each entry reviews the relevant chemistry history and uses of the compound with discussions of the origin of the compound s name the discovery or first synthesis of the compound production statistics and uses of the compound     The Naming and Indexing of Chemical Compounds from Chemical Abstracts. - Chemical

abstracts,1962     THE NAMING AND INDEXING OF CHEMICAL COMPOUNDS BY CHEMICAL ABSTRACTS. ,1945

**Rules for Nomenclature of Chemical Compounds** International Union of Pure and Applied Chemistry,1934

## Unveiling the Magic of Words: A Overview of "**Cheat Sheet For Naming Compounds**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Cheat Sheet For Naming Compounds**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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