



Automobile Engineering Interview Questions And Answers Guide.

Question - 1:

Define Fit, and Ra?

Ans:

Fit is defined as the degree of tightness and looseness between two meeting parts.

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Question - 2:

What is the working principle of hydrostatic propulsion system in a tractor?

Ans:

It depends on Pascal law. As the pressure in the system is the same, the force that the fluid gives to the surroundings is therefore equal to pressure \times area. In such a way, a small piston feels a small force and a large piston feels a large force.

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Question - 3:

WHY THE COLOR OF VEHICLE TYRES ARE ALWAYS BLACK IN COLOUR. IS THERE ANY THING RELATED WITH HEAT CONDUCTION?

Ans:

During the process of vulcanization of rubber to make it more resistant to make tyres out of it, a certain proportion of carbon is mixed into it. Without this procedure the tyres will not be able to withstand the friction & stress of the road. This carbon is responsible for giving tyres the characteristic black colour. When carbon combines with ultraviolet light from the sun, it attacks the tire polymer. To prevent this from rapidly deteriorating, a sacrificial absorber is used, "carbon black" - which, strangely, makes tires black.

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Question - 4:

What is starting off grade SAE nos of gear box in passenger car?

Ans:

"SAE" STAND "SOCIETY OF AUTOMOTIVE ENGINEERS" (USA).
SAE40W100-DRIVER ENGINE,2-WHEELER ENGINE, GEAR BOX & CLUTCH.
SAE40W100-GEAR BOX OF RWD & 4WD VEHICLE.

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Question - 5:

WHAT IS TRANSVERSE WEIGHT SHFT DUE TO DRIVE TORQUE?

Ans:

It is weight of drive torque which is added to chassis of the car.

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Question - 6:

Why tractor rear wheel is big than front wheel?

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3 lewis concept H^+ Lewis 1923 proposed the concept of acids and bases upon the electronic theory of valency to include those reactions where no proton transfer takes place according to this concept an acid

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web arrhenius bases an arrhenius base is defined as any species that increases the concentration of hydroxide ions OH^- in aqueous solution an example of an arrhenius base is the highly soluble sodium hydroxide NaOH sodium hydroxide dissociates in water as follows $\text{NaOH} \rightarrow \text{Na}^+ + \text{OH}^-$ in water sodium hydroxide

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web work done by svante arrhenius in the late 19th century set the groundwork for our current understanding of acid base theory despite the relatively limited technology available to him he was able to develop theories that adequately explained the properties and chemical reactivity of many acids and bases

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web the arrhenius theory of acids and bases acids are substances that produce hydrogen ions in solution bases are substances that produce hydroxide ions in solution neutralization happens because hydrogen ions and hydroxide ions react to produce water $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$ $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$ hydrochloric

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web aug 15 2020 the equilibrium formulated in scheme 1 between hydrogen ion and the corresponding acid and base can be called a simple acid base equilibrium by mixing two simple systems a double acid base system and an acid base equilibrium result that can always be formulated as follows $\text{acid}_1 + \text{base}_2 \rightleftharpoons \text{acid}_2 + \text{base}_1$

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web key points a brønsted lowry acid is any species that is capable of donating a proton h a brønsted lowry base is any species that is capable of accepting a proton which requires a lone pair of electrons to bond to the h water is amphoteric which means it can act as both a brønsted lowry acid and a brønsted lowry base

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web the a ion is a base because it is accepting a proton from the h 3 o the reversible reaction contains two acids and two bases we think of them in pairs called conjugate pairs when the acid ha loses a proton it forms a base a when the base a accepts a proton back again it obviously reforms the acid ha

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because it lends itself well to studying equilibrium and kine