

# Brain Pop Ions Quiz

1. Which of the following is a true statement about atomic nuclei? (Atomic nucleus)

- a. They are made up of protons, neutrons, and electrons
- b. They have a net positive charge.**
- c. Every atomic nucleus has the same number of particles
- d. The atomic nucleus has no relation to an atom's mass

2. How do the inner electron shells of an atom differ from the outer electron shells?

- a. The electrons in the outer shells orbit the nucleus at slower speeds
- b. The electrons in the outer shells have negative charges; the electrons in the inner shells have positive charges
- c. Electrons in the inner shells are gained and lost more frequently than electrons in the outer shells
- d. The outer shells can hold more electrons than the inner shells**

3. An atom with one electron in its valence shell will tend to: (Atom with one electron in valence shell)

- a. Become positively charged**
- b. Become negatively charged
- c. Accept electrons from other atoms
- d. Accept protons from other atoms

4. Cations tend to have: a. High valence b. Neutral valence **c. Positive valence** d. Negative valence

5. Fluorine is an element located on the right side of the periodic table of elements. What can you conclude about fluorine from this fact?

- a. its valence shell is nearly full**
- b. It tends to donate electrons to other atoms
- c. It tends to form cations
- d. It tends to have a positive valence

6. What can you infer about elements located in the middle of the periodic table?

- a. They tend to have very large atomic masses
- b. They tend to have completely full valence shells
- c. They tend to have positive charges
- d. They tend to form ions less frequently than elements on the edges of the table**

7. Which of the following is a true statement about ions?

- a. Removing an electron from an atom requires energy**
- b. When an atom gains an electron, it loses energy
- c. No energy is transferred when an atom gains or loses an electron
- d. Gaining and losing protons requires less energy than gaining and losing electrons

8. Which statement best describes why sodium and chlorine form bonds so frequently?

- a. A closed system can neither gain nor lose energy
- b. Opposites attract**
- c. Objects in motion tend to remain in motion
- d. What goes up must come down

9. Water is sometimes described as the "universal solvent." From the information presented in the movie, what can you infer about this nickname?

- a. It means that water can make dangerous substances harmless
- b. It means that water dissolves electrons
- c. It means that water usually has a neutral charge
- d. It means that water can break the bonds that hold many substances together**

10. How is table salt (sodium chloride) different than the ions that make it up?

- a. It has a net negative charge
- b. It has a net positive charge
- c. It has a neutral charge**
- d. It contains no electrons

# Brainpop Chemical Bonds Quiz Answers

**NARAYAN CHANGDER**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

## **Brainpop Chemical Bonds Quiz Answers:**

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