

PVS THE TUTOR ZAMBIA

CHEMISTRY WORK

SOLUTIONS

TOPICS (MATTER AND PERIODIC TABLE)

QUESTION 1

- (a) (i) A = Evaporation (the change from liquid to gas)
- (ii) B = Freezing or solidification (change from liquid to solid)
- (b) The particles in a gas move randomly from one part of the container to the other.
- (c) Change A is endothermic. This is because during evaporation heated is needed to heat up the particles for collision to take place so that more spaces are created

QUESTION 2

- (a) (i) Group I (note: only group I elements have the properties mentioned in the question.)
- (ii) $2\text{Bo}_{(s)} + 2\text{H}_2\text{O}_{(l)} \longrightarrow 2\text{Bo}(\text{OH})_{(aq)} + \text{H}_2$
- (b) Bodium would not decompose. Because group I metals require more heat to decompose in order to break the bonds in a carbonate (except lithium)

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