

AP CHEM Unit 4

UNIT 4 HOMEWORK KEY

Part 1: Chemical Reactions

1. Identify the reactants and products in the chemical equation below. Indicate the states of matter for each.

$\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$

2. Predict the products of the following reactions. Indicate the states of matter for each.

$\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$

3. Write and balance the following equations.

$\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$

4. Predict the products of the following reactions. You don't need to add coefficients.

$\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$
 $\text{Fe}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{Fe}(\text{s}) + \text{H}_2(\text{g})$

5. How can you identify a **SALT** by looking at its formula?

6. How can you identify a **MOLECULE** by looking at its formula?

Answer Keys are DETAILED!

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