

Name Key Class _____ Date _____

Nuclear Decay

Types of Radiation

1. Complete the table below to compare the properties of alpha, beta, positron, and gamma radiation.

Name	Alpha	Beta	Positron	Gamma
Greek letter	α	β^-	β^+	γ
Symbol	${}^4_2\text{He}$	${}^0_{-1}e$	${}^0_{+1}e$	N/A
Composition	$2p^+$, $2n^0$	electron	anti-electron	EM wave
Charge	$2+$	$1-$	$1+$	0
Stopped by	paper, skin, clothing	wood, glass	wood, glass	thick concrete or lead

Nuclear Decay Reactions

In problems 2 – 5, fill in the missing particle for each nuclear decay reaction. Then, match each equation to the appropriate type of nuclear decay.



A. alpha emission



B. beta emission



C. electron capture



D. positron emission

In problems 6 and 7, write a balanced equation for each nuclear decay reaction.

6. Decay of radium-223 by alpha (α) emission.



7. Decay of iodine-126 by positron (β^+) emission.



8. How are the atomic number and mass number of a nuclide affected by positron emission?

Atomic number decreases by one. Mass number stays the same.

9. How are the atomic number and mass number of a nuclide affected by the emission of gamma radiation?

Both atomic number and mass number stay the same since gamma radiation has no mass or charge.

10. Which has a slightly greater mass in a nuclear reaction, the reactants or the products? Why?

The reactants have a slightly greater mass. In a nuclear reaction, a small amount of mass is converted to energy according to the equation $E = mc^2$. The difference in mass is referred to as the mass defect.

Chemistry Worksheet Questions On Radioactivity Answers

S Marginson



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CBSE - ICSE Chemistry Part I Chandan Sengupta, First Publication October 2021 Place of Publication Arabinda Nagar Bankura 722101 This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry Some of the worksheets are prepared along with model answers Some other worksheets are meant for self assessment and evaluation purposes It is also observed that some of the topics are specific to the referred curriculum Some other topics are varyingly incorporated in other streams of study Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations

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