

Name: _____

Date: _____

PHASE CHANGE WORKSHEET

On the line at the left, write the term that best matches each description below:

condensation
phase change
volatile

vaporization
melting
equilibrium vapor pressure

melting point
sublimation
deposition

_____ sublimation

1) conversion of solid directly into a gas

_____ condensation

2) opposite of vaporization

_____ melting point

3) temperature at which solid and liquid phases exist in equilibrium.

_____ phase change

4) conversion of a substance from one of the three states of matter to another

_____ vaporization

5) change from a liquid to gas

_____ equilibrium vapor pressure

6) pressure exerted by a constant number of gas molecules in equilibrium with their liquid phase

_____ volatile

7) description of a liquid that evaporates easily, with a low boiling point and high vapor pressure

_____ deposition

8) transformation of a gas directly into a solid

_____ melting

9) phase change from a solid to a liquid

Chemistry Study Guide Phase Changes Answers

Mark Steyvers

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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William H. Myers,William T. Mooney,1980 **Prentice Hall Chemistry** ,2000 Study Guide for Mortimer's Chemistry,

Sixth Edition Donald W. Shive,Louise E. Shive,1986 **Study Guide, Introductory Chemistry, a Foundation ;**

Introductory Chemistry ; Basic Chemistry Iris Stovall,Donald J. DeCoste,2000 **Resources for Teaching Middle**

School Science Smithsonian Institution,National Academy of Engineering,National Science Resources Center of the National Academy of Sciences,Institute of Medicine,1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand

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