

NAVODAYA VIDYALAYA SAMITI.
CHEMISTRY REVISION PAPER - 4
THE 'p' BLOCK ELEMENTS

Class : XII

Max. Marks: 70

Time : 3hrs.

General Instructions:

- (i) All questions are compulsory.
- (ii) Marks for each question are indicated against it.
- (iii) Q.No.1 and 8 carrying 1 mark each.
- (iv) Q.No.9 to 18 carrying 2 marks each.
- (v) Q.No.19 to 27 carrying 3 marks each.
- (vi) Q.No. 28 to 30 carrying 5 marks each.
- (vii) Use log Tables, if necessary. Use of calculators is not permitted.

Answer the following questions:

1. Give the Resonating structures of NO_2 and N_2O_5 ?
2. How ammonia does reacts with a solution of Cu^{2+} ?
3. Does the hydrolysis of XeF_6 lead to a redox reaction?
4. Give two reasons for the anomalous behaviour of fluorine?
5. Why is $K_2 \ll K_1$ for H_2SO_4 in water?
6. Can PCl_5 act as an oxidizing as well as reducing agent? Why?
7. Give the disproportionation reaction of H_3PO_3 ?
8. PH_3 has lower boiling point than NH_3 . Why?
9. (a) Why does the S_2 molecule in vapour state shows paramagnetic behaviour?
(b) How O_3 is used in the quantitative method for the estimation of $\text{O}_{2(g)}$?
10. Explain how the five bonds in PCl_5 molecule are different in nature?
11. (a) Write the conditions to maximize the yield of H_2SO_4 by contact process.
(b) Give the reaction between SO_2 and Fe^{3+} ions and indicate whether SO_2 is oxidizing or reducing agent?
12. Comment on the nature of the two S—O bonds formed in SO_2 molecule. Indicate whether the two S—O bonds in the molecule equal or not?
13. Although electron gain enthalpy of fluorine is less negative as compared to chlorine, fluorine is a stronger oxidizing agent than chlorine. Why?
14. How can you prepare Cl_2 from HCl and HCl from Cl_2 ? Write reactions only.
15. (a) Give two reasons for the inert nature of N_2 molecule?
(b) Nitrogen cannot form $d\pi-p\pi$ bonds why?
16. Give the equations involved in the preparation of HNO_3 by Ostwald's process?
17. (a) In what way it can be proved that PH_3 is basic in nature?
(b) Which compound used in smoke screens (Holme's signals)?
18. (a) In 16th group hydrides the acidic nature decreases as we move down in the Group why?
(b) The HNH angle values is higher than HPH, HAsH and HSbH angles. Why?
19. (a) Calculate the overall complex dissociation equilibrium constant for the $\text{Cu}(\text{NH}_3)_4^{2+}$.

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Jessica J Manson



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Table of Contents Chemistry Test Sections 1and 12

1. Understanding the eBook Chemistry Test Sections 1and 12
 - The Rise of Digital Reading Chemistry Test Sections 1and 12
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Test Sections 1and 12
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Test Sections 1and 12
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Test Sections 1and 12

-
- Personalized Recommendations
 - Chemistry Test Sections 1 and 12 User Reviews and Ratings
 - Chemistry Test Sections 1 and 12 and Bestseller Lists
5. Accessing Chemistry Test Sections 1 and 12 Free and Paid eBooks
 - Chemistry Test Sections 1 and 12 Public Domain eBooks
 - Chemistry Test Sections 1 and 12 eBook Subscription Services
 - Chemistry Test Sections 1 and 12 Budget-Friendly Options
 6. Navigating Chemistry Test Sections 1 and 12 eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Test Sections 1 and 12 Compatibility with Devices
 - Chemistry Test Sections 1 and 12 Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Test Sections 1 and 12
 - Highlighting and Note-Taking Chemistry Test Sections 1 and 12
 - Interactive Elements Chemistry Test Sections 1 and 12
 8. Staying Engaged with Chemistry Test Sections 1 and 12
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Test Sections 1 and 12
 9. Balancing eBooks and Physical Books Chemistry Test Sections 1 and 12
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Test Sections 1 and 12
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Test Sections 1 and 12
 - Setting Reading Goals Chemistry Test Sections 1 and 12
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Test Sections 1 and 12

-
- Fact-Checking eBook Content of Chemistry Test Sections 1and 12
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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