Unit 5 - Counting Particles - Objectives

Re	view Concepts	
	 Types of substances Chemical formulas of substances (U 4) 	
1.	State evidence for Avogadro's Hypothesis. Use Avogadro's Hypothesis and experimental data to determine the relative mass of molecules.	
2.	Use experimental data to determine the relative mass of two objects.	
3.	Use experimental data to determine the number of items in a sample without actually counting them.	
4.	Given the chemical formula of a substance, determine the molar mass.	
5.	Given the mass of a substance, determine a. the number of moles of the sample b. the number of atoms or molecules in the sample	

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