BALANCING CHEMICAL EQUATIONS

Name GROUP

Rewrite and balance the equations below.

1. N2 + H3 - NH3 NZ + 3H2 - 2NH3

2. KClO<sub>3</sub> → KCl + O<sub>2</sub> 2 KCl O<sub>3</sub> → 2 KCl + 3 O<sub>2</sub>

3. NaCl + F2 -> NaF + Cl2 2NaCl + F2 -> ZNaF + Cl2

4. H, +0, - HO 2H2+ 02 -> 2H20

5. AgNO<sub>3</sub> + MgCl<sub>2</sub> → AgCl + Mg(NO<sub>3</sub>)<sub>2</sub> 2AgNO<sub>3</sub> + MgCl<sub>2</sub> → ZAgOl+

6. AIBr3 + K2SO4 → KBr + AL(SO4)3 2 ALBr3+ 3 K2SO4 → 6 KBr + Al2(SO4)3

7. CH, + O2 + CO2 + HO CH4+ 202 - CO2+ 2H20

8. C3H8 + O2 → CO2 + H2O C3H8+ 5O2 → 3CO2+4H2O

9. CoH10 + O2 → CO2 + HO CoH10 += 02 -> 8CO2 + 9H2O

10. FOCI, + NOOH → FO(OH), + NOCI\_ FECEX + 3 NOOH → FECOH), +3NOCE

11. P+O2 - P2O6 4P+502 -> 2P205

12. Na + H2O → NaOH + H2 2Na+ 2H2O → 2NaOH + H2

13. Ago - Ag + O2 2Ag 20 -> 4Ag + O2

14.  $S_8 + O_2 \rightarrow SO_3$   $S_8 + 12.0 = $50_3$ 

15. CO2 + H20 → C6H12O6 + O2 6CO2 + 6H2O → C6H12O6+ 6O2

16. K + MgBr<sub>2</sub> → KBr + Mg 2 K + MgBr, → 2 KBr + Mg

17. HCI + CaCO, → CaCl, + H2O + CO, 2HCI + CaCO, → Cacl+ H1O + COL

# **Chemistry Unit 6 Reaction Equations Answers**

Lei Shi

#### **Chemistry Unit 6 Reaction Equations Answers:**

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