

Unit 6 Study Guide

TOPIC 1: REVIEW

Aq (aqueous) - dissolved in water

Solution: Homogenous mixture; solutes dissolved in solvents

Solute: Dissolved particles in a solution (i.e. NaCl)

Solvent: The dissolved medium in a solution



Saturated: Solution containing max amount of solute for a given amount of solvent at a constant temperature and pressure

TOPIC 2: SOLUBILITY

- Water molecule has $\ominus$ and $\oplus$ end
- negative attached to cation
- positive attached to anion



Ion solution: water surrounding ions

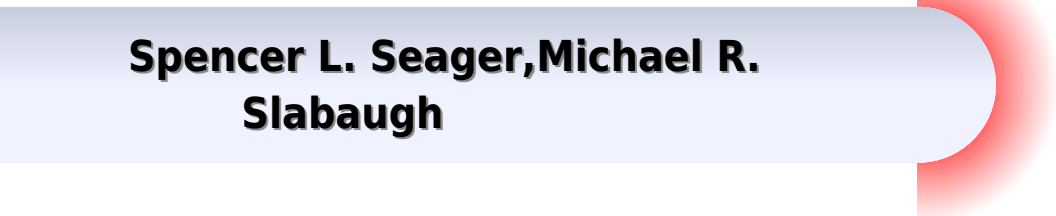
- When certain IONIC COMPOUNDS (compounds w/ 1+ atoms) are added to water, they will break apart / DISSOCIATE



Solubility rules:

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**Spencer L. Seager, Michael R.
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