

CHEMISTRY Terminology

The ABCs of chemistry

A

A: Symbol for atomic mass number
absolute zero: Lowest temperature; 0 Kelvin
Ae: Symbol for the element actinium, $Z = 89$
accuracy: The closeness of measurement to true value
acetate anion: $C_2H_3O_2^-$, charge -1
acetic acid: $C_2H_4O_2$
acetone: Dimethyl ketone; organic solvent
acid-base indicator: A chemical that changes color if the pH changes
acid-base titration: A method for determining acid or base concentrations
acid definition: See Arrhenius acid, Brønsted-Lowry acid, Lewis acid
acid ionization constant (K_a): Equilibrium constant for acid dissociation
acidic solution: pH below 7
actinide: Element with $Z = 90$ or above; radioactive
activation energy (E_a): A process's energy barrier



adhesion: Attraction of unlike molecules or materials
Ag: Symbol for the element silver, $Z = 47$
Al: Symbol for the element aluminum, $Z = 13$
alcohol: Organic compound with -OH group; ROH
aldehyde: Carbonyl group bonded to hydrogen and an organic group
alkali metal family: Lithium, sodium, potassium, rubidium, cesium, and francium; column #1
alkaline earth metal family: Beryllium, magnesium, calcium, strontium, barium, and radium; column #2
alkane: Hydrocarbon; all C-C single bonds
alkene: Hydrocarbon; 1 or more C=C double bonds
alkyne: Hydrocarbon; 1 or more C≡C triple bonds
allotropes: Two or more forms of an element
alloy: Solution of 2 or more metals
alpha (α): Greek letter that denotes radioactive particle and various scientific variables
alpha particle: Helium nucleus; charge +2
Au: Symbol for the element auriferous, $Z = 79$
amalgam: Alloy of mercury with other metal
amine: Organic base; RRRN; modified ammonia
amino acid: Compound with organic acid and organic base property; forms proteins and peptides
ammonia: NH_3 , base
ammonium cation: NH_4^+ , charge +1
ammonium hydroxide: NH_4OH
amphoteric oxide: Exhibits both acid and basic properties
amu: Atomic mass unit; $1/12$ mass of C-12
anion: Ion with a negative charge
anode: Electrode that supports oxidation
antibonding MO: MO is less stable than separate AOs
AO: Abbreviation for atomic orbital

aq: Aqueous; "dissolved in water"
Ar: Symbol for the element argon, $Z = 18$
aromatic: Organic compound with a benzene ring
Arrhenius acid: Produces hydronium ion in water solution
Arrhenius base: Yields hydroxide ion in water solution
Arrhenius equation for rate constant (k):

$$k = Ae^{-E_a/RT}$$
 E_a = activation energy
arsenide anion: As^{3-} , charge -3
As: Symbol for the element arsenic, $Z = 33$
At: Symbol for the element astatine, $Z = 85$
atm: Symbol for pressure in "atmospheres"
atom: Fundamental unit of all matter
atomic mass number (A): Total number of protons and neutrons
atomic number (Z): Number of protons in the nucleus
atomic orbital: Wave motion of electrons in atoms
atomic radius: Empirical measure of atom size
atomic weight: Weighted average of natural isotopes of an element
Au: Symbol for the element gold, $Z = 79$
Aufbau principle: Guides the filling of electronic subshells of the elements
Avogadro's law: Volume is proportional to gas moles at fixed pressure and temperature
Avogadro's number: $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$

Bk: Symbol for the element berkelium, $Z = 97$
bleach: Chemical that is a strong oxidizing agent
boiling: Liquid \rightarrow gas at the boiling point
boiling point elevation: A solution has a higher boiling point than a pure solvent
boiling point (T_b): Liquid-gas equilibrium, $P = 1 \text{ atm}$
boiling point of water: 100°C
bond energy: Energy held by a chemical bond
bond length: Distance between 2 atoms in a bond
bond order for Lewis model: # of bonds divided by # of bonded atoms
bond order for MO treatment: # of filled bonding MOs minus # of filled antibonding MOs
bonding electrons: Form chemical bonds in a compound
bonding MO: MO is more stable than AOs
Boyle's law: PV constant for gas, fixed T and n
Br: Symbol for the element bromine, $Z = 35$
bromide anion: Br^- , charge -1
Brønsted-Lowry acid: Proton donor
Brønsted-Lowry base: Proton acceptor
buffer: Composed of weak acid and weak base; serves to keep pH constant



B

B: Symbol for the element boron, $Z = 5$
Ba: Symbol for the element barium, $Z = 56$
balanced equation: Each side of the equation has the same number of atoms of each element and charge
balanced redox: Electron loss = electron gain
barium cation: Ba^{2+} , charge +2
base definition: See Arrhenius base, Brønsted-Lowry base, Lewis base
base ionization constant (K_b): Equilibrium constant for base dissociation
basic solution: pH over 7
battery: Source of electrical power; galvanic cell
Be: Symbol for the element beryllium, $Z = 4$
benzene: C_6H_6 ; organic solvent; aromatic ring of 6 carbon atoms
beryllium cation: Be^{2+} , charge +2
beta (β): Greek letter that denotes radioactive particle and a number of scientific variables
beta particle: Energetic electron from the nucleus; charge -1
Bh: Symbol for the element bohrium, $Z = 107$
Bi: Symbol for the element bismuth, $Z = 83$
bicarbonate anion: HCO_3^- , charge -1
bimolecular oxygen: O_2
binary compound: Formed from 2 elements
bisulfate anion: HSO_4^- , charge -1



b: Symbol for centi- , the SI prefix for $1/100$ or 10^{-2}
c: Symbol for the speed of light
 $^\circ\text{C}$: Celsius temperature scale
C: Symbol for the element carbon, $Z = 6$
 $C_2H_4O_2$: Acetic acid
Ca: Symbol for the element calcium, $Z = 20$
calcium cation: Ca^{2+} , charge +2
calcium fluoride: CaF_2
calcium hydroxide: $Ca(OH)_2$
calorie: English unit of heat energy
calorimetry: Study of the release or absorption of heat
carbide anion: C^{4-} , charge -4
carbohydrate: Organic compound; carbon bonded to several -OH groups (sugar and starch)
carbon dioxide: CO_2
carbon monoxide: CO
carbonate anion: CO_3^{2-} , charge -2
carbonic acid: H_2CO_3
carbonyl group: CO , ligand complex
carboxyl group: $>CO$ bonded to 2 groups
carboxylic acid: Organic acid; $R-COOH$
catalyst: Accelerates a reaction but is not consumed in the reaction; lowers the activation energy
cathode: Electrode that supports reduction
cation: Ion with a positive charge
Cd: Symbol for the element cadmium, $Z = 48$



Chemistry Vocabulary Study Guide

G. Marc Loudon, Joseph G. Stowell



Chemistry Vocabulary Study Guide:

Chemistry Regents Vocabulary Workbook Lewis Morris, Learn the Secret to Success on the Chemistry Regents Exam Ever wonder why learning comes so easily to some people This remarkable workbook reveals a system that shows you how to learn faster easier and without frustration By mastering the hidden language of the subject and exams you will be poised to tackle the toughest of questions with ease We ve discovered that the key to success on the Chemistry Regents Exam lies with mastering the Insider s Language of the subject People who score high on their exams have a strong working vocabulary in the subject tested They know how to decode the vocabulary of the subject and use this as a model for test success People with a strong Insider s Language consistently Perform better on their Exams Learn faster and retain more information Feel more confident in their courses Perform better in upper level courses Gain more satisfaction in learning The Chemistry Regents Exam Vocabulary Workbook is different from traditional review books because it focuses on the exam s Insider s Language It is an outstanding supplement to a traditional review program It helps your preparation for the exam become easier and more efficient The strategies puzzles and questions give you enough exposure to the Insider Language to use it with confidence and make it part of your long term memory The Chemistry Regents Exam Vocabulary Workbook is an awesome tool to use before a course of study as it will help you develop a strong working Insider s Language before you even begin your review Learn the Secret to Success After nearly 20 years of teaching Lewis Morris discovered a startling fact Most students didn t struggle with the subject they struggled with the language It was never about brains or ability His students simply didn t have the knowledge of the specific language needed to succeed Through experimentation and research he discovered that for any subject there was a list of essential words that when mastered unlocked a student s ability to progress in the subject Lewis called this set of vocabulary the Insider s Words When he applied these Insider s Words the results were incredible His students began to learn with ease He was on his way to developing the landmark series of workbooks and applications to teach this Insider s Language to students around the world [GCSE Chemistry Vocabulary Workbook](#) Lewis Morris, Learn the Secret to Success on the GCSE Chemistry Exam Ever wonder why learning comes so easily to some people This remarkable workbook reveals a system that shows you how to learn faster easier and without frustration By mastering the hidden language of the subject and exams you will be poised to tackle the toughest of questions with ease We ve discovered that the key to success on the GCSE Chemistry Exam lies with mastering the Insider s Language of the subject People who score high on their exams have a strong working vocabulary in the subject tested They know how to decode the vocabulary of the subject and use this as a model for test success People with a strong Insider s Language consistently Perform better on their Exams Learn faster and retain more information Feel more confident in their courses Perform better in upper level courses Gain more satisfaction in learning The GCSE Chemistry Exam Vocabulary Workbook is different from traditional review books because it focuses on the exam s Insider s Language It is an outstanding

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A Comparative Vocabulary Study Guide Spanish to English Robert D. O'Brian,2013-09-09 The

Comparative Vocabulary Study Guide Spanish to English highlights over 7 500 selected Spanish words that have either the same spelling and meaning as their English counterparts or contain easily recognizable English meanings This guide jump starts the student into a knowledge of Spanish or English by concentrating on easily recognizable words common to these languages It teaches a basic vocabulary by word association This guide is designed after the vocabulary first method of language instruction as advocated by the author The student then is taught how to structure a particular vocabulary into meaningful sentences and concepts

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The Routledge Handbook of Vocabulary Studies Stuart Webb,2019-07-30 The Routledge Handbook of Vocabulary Studies provides a cutting edge survey of current scholarship in this area Divided into four sections which cover understanding vocabulary approaches to teaching and learning vocabulary measuring knowledge of vocabulary and key issues in teaching researching and measuring vocabulary this Handbook brings together a wide range of approaches to learning words to provide clarity on how best vocabulary might be taught and learned provides a comprehensive discussion of the key issues and challenges in vocabulary studies with research taken from the past 40 years includes chapters on both formulaic language as well as single word items features original contributions from a range of internationally renowned scholars as well as academics at the forefront of innovative research The Routledge Handbook of Vocabulary Studies is an essential text for those interested in teaching learning and researching vocabulary

International Baccalaureate

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General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 **Guide to the Study and Use of Reference Books**, 1908 *Guide to the Study and Use of Reference Books* Alice Bertha Kroeger, 1904 **AP Chemistry Exam Success** Lewis Morris, 2018-07-14

Learn the Secret to Success on the AP Chemistry Exam Ever wonder why learning comes so easily to some people This remarkable book reveals a system that shows you how to learn faster easier and without frustration By mastering the hidden language of the exam you will be poised to tackle the toughest of questions with ease We've discovered that the key to success on the AP Chemistry Exam lies with mastering the Insider's Language of the subject People who score high on their exams have a strong working vocabulary in the subject tested They know how to decode the exam vocabulary and use this as a model for test success People with a strong Chemistry Insider's Language consistently Perform better in the AP Chemistry Exam Learn faster and retain more information Feel more confident in their preparation Perform better on the job Gain more satisfaction in learning The AP Chemistry Exam Success Guide is different from traditional review books because it focuses on the exam's Insider's Language It is an outstanding supplement to a traditional review program It helps your preparation for the exam become easier and more efficient The strategies puzzles and questions give you enough exposure to the Insider Language to use it with confidence and make it part of your long term memory The AP Chemistry Exam Success Guide is an awesome tool to use before a course of study as it will help you develop a strong working Insider's Language before you even begin your review Learn the Secret to Success on the AP Chemistry Exam After nearly 20 years of teaching Lewis Morris discovered a startling fact Most students didn't struggle with the subject they struggled with the language It was never about brains or ability His students simply didn't have the knowledge of the specific language needed to succeed Through experimentation and research he discovered that for any subject there was a list of essential words that when mastered unlocked a student's ability to progress in the subject Lewis called this set of vocabulary the Insider's Words When he applied these Insider's Words the results were incredible His students began to learn with ease He was on his way to developing the landmark series of Books and applications to teach this Insider's Language to students around the world Our books and applications are helpful to any student They are especially helpful to struggling students English language learners and students beginning a course of study The strongest students will also enjoy the puzzle and game aspect of the books In all cases the books provide an enjoyable break from the tedious and mundane experience of traditional test preparation Get your copy today N B When viewing our workbooks on a digital device such as a Kindle we highly recommend the use of a PDF mark up software such as Squid as it will make the experience much more effective and enjoyable *Study Guide & Solutions Manual to Accompany Organic Chemistry, Third Edition* G. Marc Loudon, Joseph G. Stowell, 1995 *A Study Guide to Chemical Principles* Wilbert Hutton, 1970 **Chemistry Exam Success** Lewis Morris, 2018-07-14

Learn the Secret to Success on the Chemistry Course and Exams Ever wonder why

learning comes so easily to some people This remarkable book reveals a system that shows you how to learn faster easier and without frustration By mastering the hidden language of the exam you will be poised to tackle the toughest of questions with ease We ve discovered that the key to success on the Chemistry Course and Exams lies with mastering the Insider s Language of the subject People who score high on their exams have a strong working vocabulary in the subject tested They know how to decode the exam vocabulary and use this as a model for test success People with a strong Chemistry Insider s Language consistently Perform better on their chemistry exams Learn faster and retain more information Feel more confident in their preparation Perform better in their classes Gain more satisfaction in learning The Chemistry Exam Success Guide is different from traditional review books because it focuses on the exam s Insider s Language It is an outstanding supplement to a traditional review program It helps your preparation for the exam become easier and more efficient The strategies puzzles and questions give you enough exposure to the Insider Language to use it with confidence and make it part of your long term memory The Chemistry Exam Success Guide is an awesome tool to use before a course of study as it will help you develop a strong working Insider s Language before you even begin your review Learn the Secret to Success on the Chemistry Course and Exams After nearly 20 years of teaching Lewis Morris discovered a startling fact Most students didn t struggle with the subject they struggled with the language It was never about brains or ability His students simply didn t have the knowledge of the specific language needed to succeed Through experimentation and research he discovered that for any subject there was a list of essential words that when mastered unlocked a student s ability to progress in the subject Lewis called this set of vocabulary the Insider s Words When he applied these Insider s Words the results were incredible His students began to learn with ease He was on his way to developing the landmark series of Books and applications to teach this Insider s Language to students around the world Our books and applications are helpful to any student They are especially helpful to struggling students English language learners and students beginning a course of study The strongest students will also enjoy the puzzle and game aspect of the books In all cases the books provide an enjoyable break from the tedious and mundane experience of traditional test preparation Get your copy today N B When viewing our workbooks on a digital device such as a Kindle we highly recommend the use of a PDF mark up software such as Squid as it will make the experience much more effective and enjoyable

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Unveiling the Power of Verbal Beauty: An Emotional Sojourn through **Chemistry Vocabulary Study Guide**

In a global inundated with displays and the cacophony of instantaneous transmission, the profound power and mental resonance of verbal beauty usually fade in to obscurity, eclipsed by the constant barrage of noise and distractions. Yet, located within the musical pages of **Chemistry Vocabulary Study Guide**, a captivating perform of fictional beauty that impulses with fresh thoughts, lies an unforgettable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this mesmerizing opus manuals viewers on a mental odyssey, softly exposing the latent potential and profound influence stuck within the elaborate internet of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is main subjects, dissect its captivating publishing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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