Chemistry Worksheet: Matter #1

- 1. A mixture (is/is not) a chemical combining of substances.
- In a compound the (atoms/molecules) are (chemically/physically)
 combined so that the elements that make up the compound (retain/lose)
 their identities and (do/do not) take on a new set of properties.
- The smallest identifiable unit of a compound is a(n) ______,
 which is made up of ______ which are chemically bonded.
- 4. True or False: A mixture is always made up of a combination of elements.
- In a mixture, the substances (lose/retain) their identities.
- In a mixture the substances involved (can/cannot) be separated by a simple physical process.
 In a compound the elements involved (can/cannot) be separated by a simple physical process because the elements are (physically combined/chemically bonded).
- 7. True or False: An element can be broken down into a simpler substance.
- The smallest identifiable unit of an element is a(n) ______.
- From the following list of substances, circle the ones that are elements: silver carbon dioxide wood alcohol chromium hydrogen carbon nitrogen water sugar salt gold oxygen magnesium air sulfur micket
- Explain how to separate the sugar and water in a solution of sugar and water.
- 11. How would you separate a mixture of alcohol and water?
- 12. How would you separate sand and water?

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