

TEACHING TRANSPARENCY WORKSHEET**9****Types of Matter****Use with Chapter 3,
Section 3.4**

1. Into what two broad classes can all matter be divided?

2. What is the difference between a mixture and a pure substance?

3. What is the difference between a compound and an element?

4. Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.

5. A list of compounds and elements is given below. Circle the substances that are elements.

gold	aluminum	water
sugar	oxygen	platinum
salt	chlorine	brass

6. How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?

7. Label each mixture below as either homogeneous or heterogeneous.

a. air _____	e. finger paint _____
b. clay _____	f. vinegar _____
c. homemade lemonade (with pulp) _____	g. soil _____
d. oatmeal raisin cookie _____	

8. List three methods that are commonly used to separate mixtures into their component substances.

Chemistry Teaching Transparency Master Answers

Charles H. Corwin



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Resources in Education**, 1994

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Introductory Chemistry Charles H. Corwin, 2005 For one semester courses in Basic Chemistry Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry This text is carefully crafted to help students learn chemical skills and concepts more effectively Corwin covers math and problem solving early in the text he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs

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