

# Term Exam 1 (Practice)

## Include answers and sufficient working.

1) Name 3 properties of a gas. (3 marks)

Indefinite shape, compressible, molecules moving, fast moving particles

2) Convert 2.1 bar to atm. (3 marks)

$$2.1 \text{ bar} \times \frac{1 \text{ atm}}{1.013 \text{ bar}} = 2.1 \text{ atm}$$

3) 10.0 g of elemental mercury weighs 22.0 g, what is the density of liquid mercury? (3 marks)

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

$$10 = \frac{22.0 \text{ g}}{\text{Volume}}$$

4) What are elements in row 6 called? How many are there? (2 marks)

See 100

5) Complete the following table. (10 marks)

	Carbon	Chlorine	Argon
Symbol	C	Cl	Ar
Relative atomic mass	12	35.5	40

6) Draw 3 structures of the same element with different numbers of electrons and label each. (4 marks)

See 100

7) Provide the chemical formula. (3 marks)



# Chemistry Term Test 1 Answers

**JA Banks**



## Chemistry Term Test 1 Answers:

**Lm Complete Gd to Ol Sci Chem 2e** ,      **Objective Tests in High School Subjects** George Frederick Miller,1926  
*Measurable Outcomes of Individual Laboratory Work in High School Chemistry* Ralph Ellison Horton,1928      *How to*  
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