Naming and Writing Formulas for Ionic Compounds

Write the names for the following compounds:

- KCIO₄ potassium perchlorate
- 2. Sr₃P₂ strontium phosphide
- 3. NH₄NO₂ ammonium nitrite
- 4. NaCl sodium chloride
- CoS cobalt (II) sulfide
- 6. ZnCl₂ zinc (II) chloride
- Li₃N lithium nitride
- BaSO_a barium sulfite
- 9. Co₂S₃ cobalt (III) sulfide
- 10. CsF cesium fluoride
- 11. NaNO₃ sodium nitrate

- 12. Cul₂ copper (II) iodide
- 13. TiO₂ titanium (IV) oxide
- KMnO₄ potassium permanganate
- 15. ZnCN2 zinc (II) cyanide
- 16. Al₂(CO₃)₃ aluminum carbonate
- 17. Fe₂(CO₃)₃ iron (III) carbonate
- 18. BaCrO₄ barium chromate
- 19. Pb₃(PO₄)₂ lead (II) phosphate
- NaHCO₃ sodium hydrogen carbonate or sodium bicarbonate

Write the formulas for the following compounds:

- cesium bromide CsBr
- aluminum chlorite Al(ClO₂)₃
- ammonium phosphite (NH₄)₃PO₃
- copper (II) nitrate Cu(NO₃)₂
- sodium fluoride NaF
- lead (IV) sulfate Pb(SO₄)₂
- calcium carbonate CaCO₃
- barium oxide BaO
- iron (III) sulfide Fe₂S₃
- manganese (II) chromate MnCrO₄

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