

## Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their predictable characteristic chemical behaviour. Atoms of one element cannot be transformed into atoms of another element, which demonstrates that each element's particles are made of atoms, regardless of the type of atom involved. However, when the elements react, molecules are formed. Molecules are made of atoms of the same element, or of different elements. We will now examine the patterns that exist for the ratios of atoms that form different molecules in order to describe the changes of the atoms during the reaction.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 1 oxygen
	1:2:2	1:2:2	1:2:2	1:2

particle



diagram

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 1 oxygen
	1:2:2	1:2:2	1:2:2	1:2

particle



diagram

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 1 oxygen
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particle



diagram

# Chemistry Unit 6 Lesson 1 Answers Key

**D Siedentop**



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