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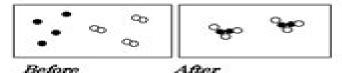
Chemistry - Balancing Equations

Balance the following equations by inserting the proper coefficients. For selected reactions, draw *Before* and *After* particle diagrams to show the particles involved in the reaction. Be sure to provide a key.

10.3

#18

- 1. $\underline{1}$ C + $\underline{1}$ H₂O \rightarrow $\underline{1}$ CO + $\underline{1}$ H₂
- 2. $2 \text{ MgO} \rightarrow 2 \text{ Mg} + 1 \text{ O}_2$
- 3. _4_Al + _3_O₂ → _2_Al₂O₃



- 4. $1 \text{ Zn} + 1 \text{ H}_2\text{SO}_4 \rightarrow 1 \text{ ZnSO}_4 + 1 \text{ H}_2$
- 5. $1 \text{ Cl}_2 + 2 \text{ KI} \rightarrow 2 \text{ KCl} + 1 \text{ I}_2$
- 6. 2 CuCl \rightarrow 2 Cu + 1 Cl₂
- 2 Na + 1 Cl₂ → 2 NaCl
- 8. $2\text{Al} + 6\text{HCl} \rightarrow 2\text{AlCl}_3 + 3\text{H}_2$



- 9. $2 \text{ Fe}_2\text{O}_3 \rightarrow 4 \text{ Fe} + 3 \text{ O}_2$
- 10. $\underline{4} P + \underline{5} O_2 \rightarrow \underline{2} P_2 O_5$
- 11. $\underline{1}$ Mg + $\underline{2}$ HCl \rightarrow $\underline{1}$ MgCl₂ + $\underline{1}$ H₂
- 12. $3 \text{ H}_2 + 1 \text{ N}_2 \rightarrow 2 \text{ NH}_3$
- 13. $\underline{1}$ BaCl₂ + $\underline{1}$ H₂SO₄ \rightarrow $\underline{1}$ BaSO₄ + $\underline{2}$ HCl
- 14. <u>1 CH4 + 2 O2 → 1 CO2 + 2 H2O</u>



#14 Before After

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