

Chemistry – Unit 7 Worksheet B

Write balanced chemical equations for the following reactions.

Hydrogen gas (collected in air) is also collected in (about) 1.5 times that volume (at STP).

1. Acetylene (C_2H_2) reacts with oxygen (hydrogen + fluorine for three acetylenes, a fluorine)
 $\text{H}_2 + \text{F}_2 \rightarrow \text{HF}$

2. Calcium carbonate is decomposed (superheating for three calcium oxide and carbon dioxide)
 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

3. Barium sulfate reacts with water for three barium hydroxide
 $\text{BaSO}_4 + \text{H}_2\text{O} \rightarrow \text{Ba(OH)}_2$

4. Methanol (CH_3OH) decomposes to form methane (CH_4) and carbon monoxide
 $\text{CH}_3\text{OH} \rightarrow \text{CH}_4 + \text{CO}$

5. Zinc reacts with oxygen (O_2) to form zinc oxide (zinc + oxygen)
 $\text{Zn} + \text{O}_2 \rightarrow \text{ZnO}$

6. Calcium sulfate decomposes (zinc for three calcium oxide and sulfur dioxide)
 $\text{CaSO}_4 \rightarrow \text{CaO} + \text{SO}_2$

7. Iron reacts with sulfur gas (Fe_2S_3) to form iron (II) sulfide and hydrogen gas
 $\text{Fe} + \text{H}_2\text{S} \rightarrow \text{FeS} + \text{H}_2$

8. A nitrogen-containing carbon compound, $\text{C}_2\text{H}_5\text{N}_3$, decomposes to form ethane, C_2H_6 , and nitrogen gas
 $\text{C}_2\text{H}_5\text{N}_3 \rightarrow \text{C}_2\text{H}_6 + \text{N}_2$

9. Phosphorus, P_4O_{10} , is formed when carbon monoxide reacts with phosphorus gas
 $\text{P}_4 + \text{CO} \rightarrow \text{P}_4\text{O}_{10} + \text{C}$

Chemistry Unit 7 Ws3 Answers

S. Chand Experts

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'S. Chand Experts' text.

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