Chemistry – Unit 1 - Worksheet 6

Dimensional Analysis

Use the factor-label method to make the following conversions. Remember to use the appropriate number of sf's in your answer.

Part 1
1. 74 cm x
$$\frac{1}{100}$$
 cm $= \frac{0.74}{100}$ meters

2. 8.32 x 10-2 kg x 1000 g = 057 grams

3. 55.5 mL x 1 cm3 = 55.5 cm3

4. 0.00527 cal x 1 kilocalories =
$$5.27 \times 10^{-6}$$
 kilocalories

5.
$$9.52 \times 10^{-4} \text{ m} \times \frac{10^{6} \text{ jum}}{10^{6}} = \frac{952}{10^{6}}$$
 micrometers

7.
$$6.0 \times 10^{-1} \text{ g} \times \frac{10009}{15} = \frac{6.0 \times 10^{3} \text{ mg}}{15}$$
8. $8.34 \times 10^{-9} \text{ cg} \times \frac{19}{10009} = \frac{8.34 \times 10^{-11}}{5.0}$

8. 8.34 x 10-9 cg x
$$\frac{19}{100000} = \frac{8.34 \times 10^{-1}}{9}$$

11.
$$5 \times 10^4 \text{ mm} \times \frac{1 \text{ m}}{1000 \text{ mm}} \times \frac{1 \text{ km}}{1000 \text{ m}} = \frac{0.05 \text{ km}}{1000 \text{ m}}$$

12.
$$9.1 \times 10^{-13} \text{ kg.} \times \frac{10009}{1 \text{ Kg}} \times \frac{10^9 \text{ rg}}{1 \text{ g}} = 0.91 \text{ rg}$$

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Oscar Krisen Buros

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Pearson Chemistry Unit 5 Staley Wilbraham (Matta, Waterman), 2022

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Table of Contents Chemistry Unit Worksheet 5 Size Of Things

- 1. Understanding the eBook Chemistry Unit Worksheet 5 Size Of Things
 - The Rise of Digital Reading Chemistry Unit Worksheet 5 Size Of Things
 - o Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Unit Worksheet 5 Size Of Things
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Unit Worksheet 5 Size Of Things
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Unit Worksheet 5 Size Of Things
 - Personalized Recommendations
 - Chemistry Unit Worksheet 5 Size Of Things User Reviews and Ratings
 - Chemistry Unit Worksheet 5 Size Of Things and Bestseller Lists
- 5. Accessing Chemistry Unit Worksheet 5 Size Of Things Free and Paid eBooks
 - Chemistry Unit Worksheet 5 Size Of Things Public Domain eBooks
 - Chemistry Unit Worksheet 5 Size Of Things eBook Subscription Services
 - Chemistry Unit Worksheet 5 Size Of Things Budget-Friendly Options

- 6. Navigating Chemistry Unit Worksheet 5 Size Of Things eBook Formats
 - o ePub, PDF, MOBI, and More
 - Chemistry Unit Worksheet 5 Size Of Things Compatibility with Devices
 - Chemistry Unit Worksheet 5 Size Of Things Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Chemistry Unit Worksheet 5 Size Of Things
 - Highlighting and Note-Taking Chemistry Unit Worksheet 5 Size Of Things
 - Interactive Elements Chemistry Unit Worksheet 5 Size Of Things
- 8. Staying Engaged with Chemistry Unit Worksheet 5 Size Of Things
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Unit Worksheet 5 Size Of Things
- 9. Balancing eBooks and Physical Books Chemistry Unit Worksheet 5 Size Of Things
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Unit Worksheet 5 Size Of Things
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Unit Worksheet 5 Size Of Things
 - Setting Reading Goals Chemistry Unit Worksheet 5 Size Of Things
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Unit Worksheet 5 Size Of Things
 - Fact-Checking eBook Content of Chemistry Unit Worksheet 5 Size Of Things
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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