hemQuest 22	
	Name:
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Information: Terminology

Recall that an ionic bond results from the combination of a metal and a nonmetal. A <u>cowalent bond</u> is the type of bond between two nonmetals. Cowalent bonds are formed by neutral atoms that share electrons rather than by charged ions. When a compound is formed by sharing electrons, the compound is called a <u>molecule</u> or molecular compound. It is important to note that ionic compounds are <u>not</u> called molecules. The largest class of molecules are called <u>organic molecules</u> <u>Carbon</u> is the distinguishing mark of organic compounds.

Critical Thinking Questions

Circle any of the following compounds that would properly be called a "molecule".

a) H₂O

b) CO₂

c) NaCl

d) Mg_bP₂

e) N₂O₂

Information: Naming Covalent Compounds

There are several prefixes used to name molecules. The name "carbon oxide" is not sufficient because carbon and oxygen sometimes form CO₂ and sometimes CO. Prefixes are necessary to distinguish between them.

Formula	Name
N ₂ O ₄	dinitrogen tetraoxide
SF∉	sulfur hexafluoride
XeCls	xenon pentachloride
SO ₂	sulfur trioxide
CO	carb on monoxide

Critical Thinking Questions

Fill in the table to indicate which prefix is used to represent the numbers. The first one is done for you.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa

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M Mark

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Making the Unseen Real Mark Vincent Falzon, 2009 Chemical Week, 2000 Training and Education for Online Institute of Information Scientists, 1989 **Detroit Suburban North Woodward Area Telephone Directories**, 1994

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