

Challenging 100

Intro to Moles

Name: _____
 Date: _____
 Score: _____

Information: Thinking in Quantities

A "mole" is simply a quantity of about just like a dozen (12) eggs. If you have 1 dozen eggs and I have 1 dozen eggs, we have 24 eggs. Similarly, a chemist will use a number instead of having to say "a million molecules" or "a billion molecules". Just like the number "12" goes with "dozen," the prefix "mole" goes with "molecules" and we will worry about that later.

Below are two boxes of molecules that represent 1 mole of molecules. There are 6.022 x 10²³ molecules in 1 mole.



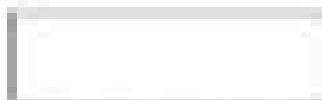
This picture represents a small number of molecules. The gas molecules are blue spheres.



This picture represents a small number of molecules. The gas molecules are blue spheres.

Critical Thinking Questions

1. Which following correctly picture 1 mole of molecules (about 6.022 x 10²³) of gas?



2. What is the mass of 1 mole of the gas in the previous question (2)?

3. How much does 1 mole of gas weigh? (2) How much does 1 mole of gas weigh? (2)



4. The picture represents 1 mole of gas. How much does 1 mole of gas weigh?

5. How much does 1 mole of gas weigh? (2) How much does 1 mole of gas weigh? (2)

Chemquest 39 Answers Intro To Moles

RJ Shavelson



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