

PERIODIC TABLE TRENDS

Figure: Ancient Egypt

Use colored pencils to color in each block and answer the questions below.

Class: _____ Date: _____
s, p, d, and f Blocks

1												18																																	
H																		He																											
Li	Be											B	C	N	O	F	Ne																												
Na	Mg											Al	Si	P	S	Cl	Ar																												
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr																												
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe																												
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn																												
Fr	Ra	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og																												
<table><tr><td>La</td><td>Ce</td><td>Pr</td><td>Nd</td><td>Pm</td><td>Sm</td><td>Eu</td><td>Gd</td><td>Tb</td><td>Dy</td><td>Ho</td><td>Er</td><td>Tm</td><td>Yb</td></tr><tr><td>Ac</td><td>Th</td><td>Pa</td><td>U</td><td>Np</td><td>Pu</td><td>Am</td><td>Cm</td><td>Bk</td><td>Cf</td><td>Es</td><td>Fm</td><td>Md</td><td>No</td></tr></table>																		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb																																
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No																																

s block: red
p block: yellow
d block: blue
f block: purple

Class: _____ Date: _____
Historical Periodic Tables

Group IV BSP BSP	Group V BSP BSP	Group VI BSP BSP	Group VII BSP BSP	Group VIII BSP BSP
Carbon	Nitrogen	Oxygen	Fluorine	Neon
Phosphorus	Sulfur	Chlorine	Bromine	Krypton
Iron	Cobalt	Nickel	Copper	Zinc
Mercury	Gold	Silver	Platinum	Palladium
Lead	Antimony	Arsenic	Germanium	Indium
Thallium	Polonium	Astatine	Radon	

A notice that there are blanks in the periodic table is that one is "G" - The is white element on diamond which element is "G" - How...

11 CHEMISTRY
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For each of the elements below, write which block they are in.

C	Cl	Cn
1. s	2. p	3. f
Cd	Cu	Cf
4. d	5. d	6. f

22. For each of the type of subshell, write the number of electrons that can fit in each subshell.

Chapter Study Guide Periodic Trends Answers

F Rizvi



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