

## Stoichiometry Review

1. The combustion of propane occurs via the reaction:



How many grams of oxygen are required to burn completely 10.0 g of propane?

2. Small quantities of chlorine can be prepared in the laboratory by the reaction:



What volume of chlorine at STP can be prepared from 100.0 g of manganese dioxide?

3. The most common ore of arsenic is mispickel,  $\text{FeSAs}$ . Upon heating this ore, free arsenic is obtained. How many grams of  $\text{FeSAs}$  are required to produce 10.0 g of arsenic according to the following reaction?



4. Glucose is used as an energy source by the human body. The overall reaction in the body is:



Calculate the volume of oxygen at STP required to convert 28.0 g of glucose to carbon dioxide and water.

5. Potassium nitrate is widely used as a fertilizer because it provides two essential elements, potassium and nitrogen. It is made by mixing potassium chloride and nitric acid in the presence of oxygen according to the equation:



How many kilograms of potassium nitrate will be produced from 50.0 kg of potassium chloride and 50.0 kg of nitric acid with excess oxygen?

6. Bromine can be prepared by adding chlorine to an aqueous solution of sodium bromide. How many grams of bromine are formed if 25.0 g of chlorine and 25.0 g of sodium bromide are reacted according to the reaction below? Which reactant is in excess and how much remains after the reaction is complete?



7. How many millilitres of 2.00 M hydrochloric acid are required to react with 2.55 g of zinc according to the following reaction:



8. A 0.473 g sample of phosphorus is reacted with an excess of chlorine. Calculate the mass of phosphorus pentachloride produced.

9. How many grams of oxygen can be prepared by the decomposition of 25.0 g of mercury(II) oxide?

10. What volume of oxygen gas at STP can be prepared by the decomposition of 25.0 g of potassium chlorate?

11. How many grams of zinc are required for the replacement of 0.100 g of hydrogen from sulfuric acid?

12. What mass of magnesium chloride is required to react with 10.0 g of silver nitrate?

# Chapter Stoichiometry Review Answers Section 2

**Thomas A. Evangelist**



## Chapter Stoichiometry Review Answers Section 2:

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## **Table of Contents Chapter Stoichiometry Review Answers Section 2**

1. Understanding the eBook Chapter Stoichiometry Review Answers Section 2
  - The Rise of Digital Reading Chapter Stoichiometry Review Answers Section 2
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Stoichiometry Review Answers Section 2
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter Stoichiometry Review Answers Section 2
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Stoichiometry Review Answers Section 2
  - Personalized Recommendations
  - Chapter Stoichiometry Review Answers Section 2 User Reviews and Ratings
  - Chapter Stoichiometry Review Answers Section 2 and Bestseller Lists

5. Accessing Chapter Stoichiometry Review Answers Section 2 Free and Paid eBooks
  - Chapter Stoichiometry Review Answers Section 2 Public Domain eBooks
  - Chapter Stoichiometry Review Answers Section 2 eBook Subscription Services
  - Chapter Stoichiometry Review Answers Section 2 Budget-Friendly Options
6. Navigating Chapter Stoichiometry Review Answers Section 2 eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter Stoichiometry Review Answers Section 2 Compatibility with Devices
  - Chapter Stoichiometry Review Answers Section 2 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter Stoichiometry Review Answers Section 2
  - Highlighting and Note-Taking Chapter Stoichiometry Review Answers Section 2
  - Interactive Elements Chapter Stoichiometry Review Answers Section 2
8. Staying Engaged with Chapter Stoichiometry Review Answers Section 2
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter Stoichiometry Review Answers Section 2
9. Balancing eBooks and Physical Books Chapter Stoichiometry Review Answers Section 2
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter Stoichiometry Review Answers Section 2
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter Stoichiometry Review Answers Section 2
  - Setting Reading Goals Chapter Stoichiometry Review Answers Section 2
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter Stoichiometry Review Answers Section 2
  - Fact-Checking eBook Content of Chapter Stoichiometry Review Answers Section 2
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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