

Stoichiometry Worksheet
Chemistry

Name _____

Date _____

1. In the reaction: $2\text{Al}_2\text{O}_3 \rightarrow 4\text{Al} + 3\text{O}_2$, the mole ratios are:

Al_2O_3 to Al _____

Al_2O_3 to O_2 _____

Al to O_2 _____

O_2 to Al _____

2. In the reaction: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$, the mole ratio of oxygen to water is: _____

3. In the reaction: $\text{Ca} + \text{Cl}_2 \rightarrow \text{CaCl}_2$, the mole ratio of chlorine to calcium is: _____

4. In the reaction: $2\text{Fe} + 6\text{HCl} \rightarrow 2\text{FeCl}_3 + 3\text{H}_2$, the mole ratio of iron to hydrochloric acid is: _____

5. The equation for the Haber process is represented by:



To convert 9.0 moles of hydrogen to ammonia would require how many moles of nitrogen?

6. In the equation: $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$, how many moles of oxygen are produced when 3.00 moles of KClO_3 decompose completely?

7. For the reaction: $\text{C}_2\text{H}_4 + 3\text{H}_2 \rightarrow \text{CH}_4$, how many moles of H_2 are required to produce 10.0 moles of CH_4 ?

Chapter Stoichiometry Worksheet

O. Jerry Parker, Gary L. Breneman

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

Chapter Stoichiometry Worksheet:

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and emphasis is placed on multi device flowsheets with recycle bypass and purge streams whose material and heat balance equations were previously too complicated to solve by the normally used hand calculator The Excel based program FlowBal helps the user set up material and heat balance equations for processes with multiple streams and units **How to Solve**

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