

NAME: \_\_\_\_\_

CLASS: \_\_\_\_\_

DATE: \_\_\_\_\_

1. The amount of thermal energy stored in an object depends on:

- |                         |   |                         |                                |
|-------------------------|---|-------------------------|--------------------------------|
| <input type="radio"/> A | All of the above  | <input type="radio"/> B | the mass of the object         |
| <input type="radio"/> C | the amount of energy that the particular material stores per degree of temperature. | <input type="radio"/> D | the temperature of the object. |

2. Heat is defined as the:

- |                         |   |                         |   |
|-------------------------|---|-------------------------|---|
| <input type="radio"/> A | average kinetic energy of the molecules of an object. | <input type="radio"/> B | average temperature of a substance.                       |
| <input type="radio"/> C | amount of thermal energy an object contains.          | <input type="radio"/> D | flow of thermal energy from one object to another object. |

3. When thermal energy flows, it raises the temperature of some substances more than others. This property of a substance is called:

- |                         |                |                         |              |
|-------------------------|----------------|-------------------------|--------------|
| <input type="radio"/> A | equilibrium.   | <input type="radio"/> B | latent heat. |
| <input type="radio"/> C | specific heat. | <input type="radio"/> D | dew point.   |

4. If you hold a paper cup containing hot chocolate in your hand, thermal energy:

- |                         |  |                         |  |
|-------------------------|--|-------------------------|--|
| <input type="radio"/> A | does not flow.                             | <input type="radio"/> B | flows from the air to the hot chocolate.   |
| <input type="radio"/> C | flows from your hand to the hot chocolate. | <input type="radio"/> D | flows from the hot chocolate to your hand. |

5. When you put ice in a drink to cool it:

- |                         |   |                         |  |
|-------------------------|---|-------------------------|--|
| <input type="radio"/> A | the ice cracks and releases cold air molecules that cool off the drink. | <input type="radio"/> B | heat from the ice is lost to the warmer liquid around it.    |
| <input type="radio"/> C | coldness is transferred from the ice to the warmer drink.               | <input type="radio"/> D | heat is transferred from the warmer drink to the cooler ice. |

# Chapter Test Thermal Energy Answers 16

**Yicheng Fang**



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