

Chapter 6

The Chemistry of Life, continued

Reinforcement and Study Guide

Section 6.2 Water and Diffusion

In your notebook, read about water and its importance.

For each statement below, write **true** or **false**.

- _____ 1. In a water molecule, electrons are shared equally between the hydrogen atoms and oxygen atom.
- _____ 2. The attraction of opposite charges between hydrogen and oxygen forms a weak oxygen bond.
- _____ 3. Because of its polarizing, water can move from the roots of a plant up to its leaves.
- _____ 4. Water changes temperature easily.
- _____ 5. Unlike most substances, water expands when it freezes.

Circle the letter of the choice that best completes the statement.

- 6. All objects in motion have
 - a. potential energy.
 - b. heat energy.
 - c. kinetic energy.
 - d. random energy.
- 7. The first scientist to observe evidence of the random motion of molecules was
 - a. Brown.
 - b. Darwin.
 - c. Mendel.
 - d. Huxley.
- 8. The net movement of particles from an area of higher concentration to an area of lower concentration is called
 - a. dynamic equilibrium.
 - b. osmosis and movement.
 - c. concentration gradient.
 - d. diffusion.
- 9. Diffusion occurs because of
 - a. osmosis and movement of particles.
 - b. random movement of particles.
 - c. a chemical reaction between particles.
 - d. chemical energy.
- 10. When a few drops of colored corn syrup are added to a beaker of pure corn syrup, the color will
 - a. move from low concentration to high concentration.
 - b. form a polar bond.
 - c. start to diffuse.
 - d. remain on the bottom of the beaker.
- 11. Diffusion can be accelerated by
 - a. decreasing the pressure.
 - b. increasing the temperature.
 - c. decreasing the movement of particles.
 - d. increasing the dynamic equilibrium.
- 12. When materials pass into and out of a cell at equal rates, there is no net change in concentration inside the cell. The cell is in a state of
 - a. dynamic equilibrium.
 - b. osmosis.
 - c. imbalance.
 - d. inertia.
- 13. The difference in concentration of a substance across space is called
 - a. dynamic equilibrium.
 - b. concentration gradient.
 - c. diffusion.
 - d. Brownian movement.

Chapter The Chemistry Of Life Reinforcement And Study Guide Answers

**Susan Hartley, David G. Myers, Steven
J. Spencer**



Chapter The Chemistry Of Life Reinforcement And Study Guide Answers:

Chemistry and Life John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993 *Fundamentals of Organic Chemistry, Textbook, Study Guide and Solutions Manual* T. W. Graham Solomons, 1996-08-15 A realistic approach to the study of mechanisms The book addresses real functional group chemistry with an emphasis on the biological environmental and medical applications of organic chemistry Organic Chemistry Marye Anne Fox, James K. Whitesell, 2004 Accompanying CD ROM has been enhanced with updated animated illustrations to accompany the presentations and Chem3D files for helpful structure visualization Page 4 of cover *Resources for Teaching Middle School Science* Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum

specialists advocates of hands on science teaching and concerned parents *Science Spectrum* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 General, Organic, and Biochemistry Media Update Ira Blei, George Odian, 2008-12-19 Blei and Odian's text gives students the tools they need to develop a working understanding of chemical principles rather than just asking them to memorize facts Now available in a new media enhanced version complete with its own online course space learning environment ChemPortal Blei Odian is better suited than ever to meet the needs of the students taking this course The Media Update version of Blei Odian includes references to dynamic interactive tutorials which provide a step by step walkthrough of concepts and problem solving skills as well as answer specific feedback and practice problems We recognize that all introductory courses are not alike For that reason we offer this text in three versions so you can choose the option that's right for you General Organic and Biochemistry cloth 0 7167 4375 2 paper 1 4292 0994 1 the comprehensive 26 chapter text An Introduction to General Chemistry 0 7167 7073 3 10 chapters that cover the core concepts in general chemistry Organic and Biochemistry 0 7167 7072 5 16 chapters that cover organic and biochemistry plus two introductory chapters that review general chemistry *Cells*, 1997 **Merrill Earth Science** Ralph M. Feather, Susan Leach Snyder, Dale T. Hesser, 1995 **Beginning and Intermediate Algebra** K. Elayn Martin-Gay, 2001 Thirteen chapters discuss real numbers equations inequalities and problem solving graphs and functions solving systems of linear equations exponents and polynomials factoring polynomials rational expressions transitions to intermediate algebra radicals rational exponents and complex numbers The Science Teacher, 1987 Resources in Education, 1984 Serves as an index to Eric reports microform **Program Descriptions for Science Instructional Materials**, 1986 **Prentice Hall Chemistry**, 2000 Study Guide for Use with Social Psychology, Second Canadian Edition Susan Hartley, David G. Myers, Steven J. Spencer, 2003-07-30 **Matter** Prentice-Hall Staff, 1994 **Science Books & Films**, 1989 **Resources in Education**, 1987 **Choice**, 1990 **Choice** Richard K. Gardner, Phyllis Grumm, 1976 **Environmental Science and Technology** Diana L. Turner, 2003

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