

Chapter 5: Thermochemistry
Problem Set with Answers

1. A system receives 425 J of heat and delivers 425 J of work to its surroundings. What is the change in internal energy of the system (in J)?

Answer: 0 J

2. Complete combustion of 1.0 metric ton of coal (assuming pure carbon) to gaseous carbon dioxide releases 3.3×10^{10} J of heat. Convert this energy to (a) kilojoules; (b) kilocalories

Answer: (a) 3.3×10^7 kJ (b) 7.9×10^6 kcal

3. Classify the following processes as exothermic or endothermic: (a) freezing of water; (b) boiling of water; (c) digestion of food; (d) a person running; (e) a person growing; (f) wood being chopped; (g) heating with a furnace.

Answer: (a) exothermic (b) endothermic (c) exothermic (d) exothermic
(e) endothermic (f) endothermic (g) exothermic

4. Calculate q when 12.0 g of water is heated from 20 °C to 100 °C.

Answer: 4.0×10^3 J

5. A 295-g aluminum engine part at an initial temperature of 3.00°C absorbs 85.0 kJ of heat. What is the final temperature of the part (c of Al = 0.900 J/g·K)?

Answer: 323°C

6. Ethylene (C_2H_4) is the starting material for the preparation of polyethylene. Although typically made during the processing of petroleum, ethylene occurs naturally as a fruit-ripening hormone and as a component of natural gas.
(a) The heat of reaction for the combustion of C_2H_4 is -1411 kJ/mole. Write a balanced thermochemical equation for the combustion of C_2H_4 .
(b) How many grams of C_2H_4 must burn to give 70.0 kJ of heat?

Answer: (a) $\text{C}_2\text{H}_4 (\text{g}) + 3\text{O}_2 (\text{g}) \rightarrow 2\text{CO}_2 (\text{g}) + 2\text{H}_2\text{O} (\text{l})$; $\Delta H_{\text{rxn}} = -1411 \text{ kJ}$
(b) 1.39 g C_2H_4

Chapter Thermochemistry Answers

James P. Birk, Donald M. Baird



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Solutions Manual to Accompany Elements of Physical Chemistry C. A. Trapp, M. P. Cady, 2007 The Solutions manual to accompany Elements of Physical Chemistry 4e contains full worked solutions to all end of chapter exercises featured in the book **Complete Preparation for the MCAT** Williams & Wilkins Review, 1998-04 Here is the most respected test prep book for the Medical College Admission Test you can buy featuring an active learning approach for a better understanding of the exam's content and a better chance for success Unique to this guide are coverage of all recent changes in the MCAT plus a step by step plan for sharpening cognitive skills developing problem solving skills and critical thinking This thorough guide replaces expensive test preparation courses while giving students exactly what they need to get ready for the MCAT

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Table of Contents Chapter Thermochemistry Answers

1. Understanding the eBook Chapter Thermochemistry Answers
 - The Rise of Digital Reading Chapter Thermochemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Thermochemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter Thermochemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Thermochemistry Answers
 - Personalized Recommendations
 - Chapter Thermochemistry Answers User Reviews and Ratings
 - Chapter Thermochemistry Answers and Bestseller Lists
5. Accessing Chapter Thermochemistry Answers Free and Paid eBooks
 - Chapter Thermochemistry Answers Public Domain eBooks
 - Chapter Thermochemistry Answers eBook Subscription Services
 - Chapter Thermochemistry Answers Budget-Friendly Options

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 - Chapter Thermochemistry Answers Compatibility with Devices
 - Chapter Thermochemistry Answers Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Chapter Thermochemistry Answers
 - Highlighting and Note-Taking Chapter Thermochemistry Answers
 - Interactive Elements Chapter Thermochemistry Answers
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter Thermochemistry Answers
9. Balancing eBooks and Physical Books Chapter Thermochemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter Thermochemistry Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter Thermochemistry Answers
 - Setting Reading Goals Chapter Thermochemistry Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter Thermochemistry Answers
 - Fact-Checking eBook Content of Chapter Thermochemistry Answers
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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