



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the name corresponding to the formula)



Directions for naming:

1. Write the name of the cation first.
2. Write the name of the anion second.
3. If the cation is a transition metal, write its name and its charge in parentheses.
4. If the anion is a polyatomic ion, write its name.
5. If the anion is a monatomic ion, write its name and its charge in parentheses.
6. If the compound is an acid, write its name.

Binary Compounds

First Element	Anion
H ₂	hydrogen
Li	lithium
Na	sodium
K	potassium
Rb	rubidium
Cs	cesium
Ba	barium
Ca	calcium
Mg	magnesium
Be	beryllium
Al	aluminum
Zn	zinc
Ag	silver
Cd	cadmium
Pb	lead
Sn	tin
Pt	platinum
Au	gold

Common Ions (Cations)

Formula	Name
H ⁺	hydrogen
Li ⁺	lithium
Na ⁺	sodium
K ⁺	potassium
Rb ⁺	rubidium
Cs ⁺	cesium
Ba ²⁺	barium
Ca ²⁺	calcium
Mg ²⁺	magnesium
Be ²⁺	beryllium
Al ³⁺	aluminum
Zn ²⁺	zinc
Ag ⁺	silver
Cd ²⁺	cadmium
Pb ²⁺	lead
Sn ²⁺	tin
Pt ²⁺	platinum
Au ³⁺	gold

Naming Compounds with a Central and Monatomic



Carbon Dioxide

Directions for naming:

1. Write the name of the cation first.
2. Write the name of the anion second.
3. If the cation is a transition metal, write its name and its charge in parentheses.
4. If the anion is a polyatomic ion, write its name.
5. If the anion is a monatomic ion, write its name and its charge in parentheses.

Common Ions (Anions)

Formula	Name	Formula	Name
F ⁻	fluoride	Br ⁻	bromide
Cl ⁻	chloride	I ⁻	iodide
S ²⁻	sulfide	At ⁻	astatine
O ²⁻	oxide		
N ³⁻	nitride		
C ⁴⁻	carbide		
B ³⁻	boride		
Al ³⁻	aluminum		
Si ⁴⁻	silicide		
P ³⁻	phosphide		
As ³⁻	arsenide		
Sb ³⁻	antimonide		
Bi ³⁻	bismuthide		
Te ²⁻	telluride		
Se ²⁻	selenide		
Mo ²⁻	molybdenide		
Cr ²⁻	chromide		
Mn ²⁻	manganese		
Fe ²⁻	iron		
Co ²⁻	cobalt		
Ni ²⁻	nickel		
Cu ²⁻	copper		
Zn ²⁻	zinc		
Ag ²⁻	silver		
Cd ²⁻	cadmium		
Pb ²⁻	lead		
Sn ²⁻	tin		
Pt ²⁻	platinum		
Au ²⁻	gold		

Common Ions for Specific Elements

Formula	Name	Formula	Name
H ⁺	hydrogen	H ⁻	hydride
Li ⁺	lithium	Li ⁻	lithide
Na ⁺	sodium	Na ⁻	sodium
K ⁺	potassium	K ⁻	potassium
Rb ⁺	rubidium	Rb ⁻	rubidium
Cs ⁺	cesium	Cs ⁻	cesium
Ba ²⁺	barium	Ba ⁻	barium
Ca ²⁺	calcium	Ca ⁻	calcium
Mg ²⁺	magnesium	Mg ⁻	magnesium
Be ²⁺	beryllium	Be ⁻	beryllium
Al ³⁺	aluminum	Al ⁻	aluminum
Zn ²⁺	zinc	Zn ⁻	zinc
Ag ⁺	silver	Ag ⁻	silver
Cd ²⁺	cadmium	Cd ⁻	cadmium
Pb ²⁺	lead	Pb ⁻	lead
Sn ²⁺	tin	Sn ⁻	tin
Pt ²⁺	platinum	Pt ⁻	platinum
Au ³⁺	gold	Au ⁻	gold

Cheat Sheet For Naming Compounds

M Walker



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abstracts,1962 **THE NAMING AND INDEXING OF CHEMICAL COMPOUNDS BY CHEMICAL ABSTRACTS.** ,1945

Rules for Nomenclature of Chemical Compounds International Union of Pure and Applied Chemistry,1934

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