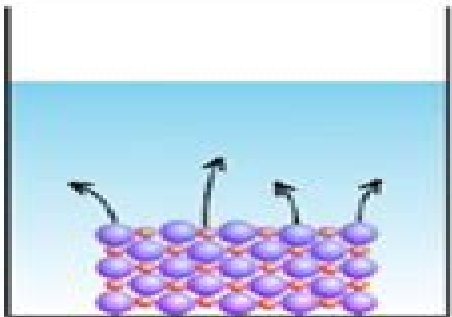
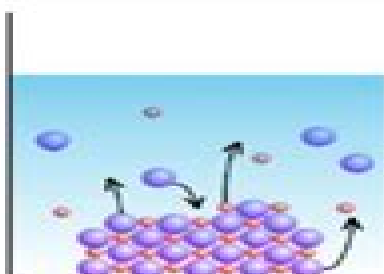
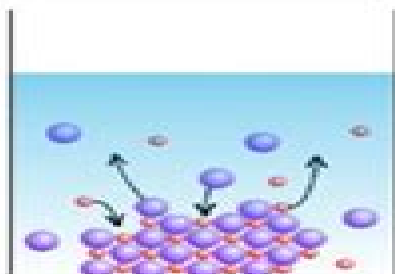


Review of Solubility Principles and Calculations

 <p>Salt is initially put into the water and begins dissolving.</p>	 <p>Salt continues to dissolve; however, dissolved ions will also precipitate. Because the salt dissolves faster than its ions precipitate, the net movement is towards dissolution.</p>	 <p>Eventually, the rate of dissolution will equal the rate of precipitation. The solution will be in equilibrium, but the ions will continue to dissolve and precipitate.</p>
--	--	---

Saturated solutions of ionic substances are shown as follows:

Remember that molecular substances do not form ions when they dissolve and thus saturated solutions of molecular substances are shown as follows:

Remember it is the ions in solution that conduct electricity, therefore molecular solutions do not conduct electricity.

Solubility Curves

Solubility curves are plots of the solubility of a substance with respect to _____. Often more than one substance is depicted on one set of axis. The units are typically given in _____. A solubility curve can provide a great deal of information including whether the dissolution of a solute is _____ or _____ and whether a solution will be saturated,

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Marcel A. Müller



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