

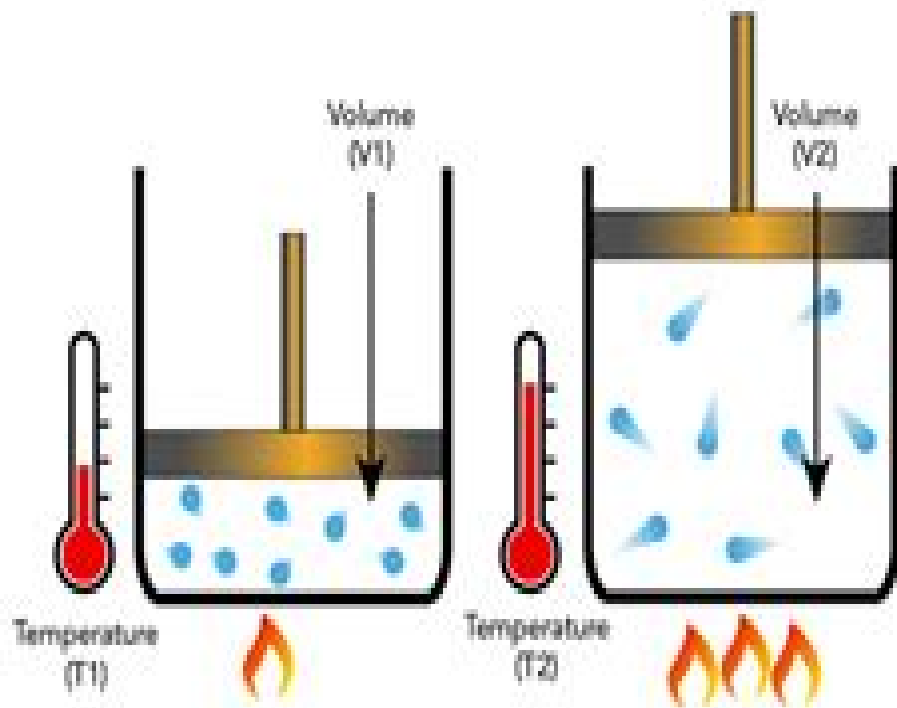
THE CHARLES'S LAW

Charles's Law, named after French scientist Jacques Charles, describes the relationship between the volume and temperature of a gas at constant pressure.

It states that the volume of a gas is directly proportional to its absolute temperature when measured in Kelvin. Mathematically, it is expressed as;

$$V_1/T_1 = V_2/T_2$$

where V_1 and T_1 are the initial volume and temperature, and V_2 and T_2 are the final volume and temperature. Charles's Law is a fundamental principle in gas physics and helps explain the behavior of gases under changing temperature conditions.



Conceptual Chemistry Charles Law Ws

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