

## Concept Review:

1. Draw the Lewis diagram for each compound.
  - A. a molecule composed of two chlorine atoms
  - B. a molecule composed of a hydrogen atom and a bromine atom

## Answers:

### Solution

Chlorine has the same valence shell electron configuration as fluorine, so the Lewis diagram for a molecule composed of two chlorine atoms is similar to the one for fluorine:



Bromine has the same valence shell electron configuration as fluorine, so the Lewis diagram for a molecule composed of a hydrogen atom and a bromine atom is similar to that for hydrogen and fluorine:



## Test Yourself:

1. Draw the Lewis diagram for each compound.
  - A. a molecule composed of one chlorine atom and one fluorine atom
  - B. a molecule composed of one hydrogen atom and one iodine atom

# Covalent Bond Concept Review

**Leo J. Malone**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## Covalent Bond Concept Review:

**Concepts in Biology' 2007 Ed.2007 Edition ,      Fundamentals of Microbiology** Jeffrey C. Pommerville,2014-12  
Ideal for health science and nursing students Fundamentals of Microbiology Body Systems Edition Third Edition retains the engaging student friendly style and active learning approach for which award winning author and educator Jeffrey Pommerville is known Highly suitable for non science majors the fully revised and updated third edition of this bestselling text contains new pedagogical elements and an established learning design format that improves comprehension and retention and makes learning more enjoyable Unlike other texts in the field Fundamentals of Microbiology Body Systems Edition takes a global perspective on microbiology and infectious disease and supports students in self evaluation and concept absorption Furthermore it includes real life examples to help students understand the significance of a concept and its application in today s world whether to their local community or beyond New information pertinent to nursing and health sciences has been added while many figures and tables have been updated revised and or reorganized for clarity Comprehensive yet accessible the Third Edition is an essential text for non science majors in health science and nursing programs taking an introductory microbiology course Provided by publisher      **Principles of Life** David M. Hillis,David E. Sadava,Richard Hill,Mary V. Price,2014-07-15 With its first edition Principles of Life provided a textbook well aligned with the recommendations proposed in BIO 2010 Transforming Undergraduate Education for Future Research Biologists and Vision and Change in Undergraduate Biology Education Now Principles of Life returns in a thoroughly updated new edition that exemplifies the reform that is remaking the modern biology classroom      **Ebook: Chemistry: The Molecular Nature of Matter and Change** Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change

**Loose-leaf Version for Principles of Life** David M. Hillis,David E. Sadava,Richard W. Hill,Mary V. Price,2013-12-01 With its first edition Principles of Life provided a textbook well aligned with the recommendations proposed in BIO 2010 Transforming Undergraduate Education for Future Research Biologists and Vision and Change in Undergraduate Biology Education Now Principles of Life returns in a thoroughly updated new edition that exemplifies the reform that is remaking the modern biology classroom      *Basic Concepts of Chemistry, Study Guide* Leo J. Malone,2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems      *Chemical Interactions* ,2005      *Human Biology* Daniel Chiras,2012 Written for the introductory human biology course the Seventh Edition of Chiras acclaimed text maintains the original

organizational theme of homeostasis presented in previous editions to present the fundamental concepts of mammalian biology and human structure and function Chiras discusses the scientific process in a thought provoking way that asks students to become deeper more critical thinkers The focus on health and homeostasis allows students to learn key concepts while also assessing their own health needs An updated and enhanced ancillary package includes numerous student and instructor tools to help students get the most out of their course

**Chemistry3** Andrew Burrows, John Holman, Andrew Parsons, Gwen Pilling, Gareth Price, 2017 Chemistry3 establishes the fundamental principles of all three strands of chemistry organic inorganic and physical By building on what students have learned at school using carefully worded explanations annotated diagrams and worked examples it presents an approachable introduction to chemistry and its relevance to everyday life

Holt Chemistry Holt Rinehart & Winston, 2003-01-24

Human Biology Chiras, 2018-02-16 Dan Chiras once again offers a refreshing and student friendly introduction to the structure function health and homeostasis of the human body in a modernized ninth edition of Human Biology This acclaimed text explores life from a variety of levels and perspectives including cellular molecular by body system through disease and within the environment

Chemical Engineering and Mining Review , 1927

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

**Fundamentals of Microbiology** Jeffrey C. Pommerville, 2017-05-02 Pommerville s Fundamentals of Microbiology Eleventh Edition makes the difficult yet essential concepts of microbiology accessible and engaging for students initial introduction to this exciting science

**Student Solutions Manual to Accompany Chemistry & Chemical Reactivity, Fourth Edition, Kotz & Treichel** Alton J. Banks, Paul Treichel, 1999

**Polymer Journal** , 2006

Concepts in Biochemistry William Kay Stephenson, 1967

**Annual Review of Psychology** Calvin Perry Stone, 1981 Publishes original critical reviews of the significant literature and current developments in psychology

Annual Review of Psychology Mark R. Rosenzweig, Lyman W. Porter, 1980

The Electron-repulsion Theory of the Chemical Bond William Fay Luder, 1967

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