

Name: Answer Key Date: _____ Per.: _____

Covalent Bond Practice

1. Fill in the missing information on the chart.

Element	# of Protons	# of Electrons	# of Valence Electrons	# of electrons to fill outer shell.
Carbon	6	6	4	4
Hydrogen	1	1	1	1
Chlorine	17	17	7	1
Helium	2	2	2	0
Phosphorus	15	15	5	3
Oxygen	8	8	6	2
Sulfur	16	16	6	2
Nitrogen	7	7	5	3

2. For each of the following covalent bonds:

- Write the symbols for each element.
- Draw a Lewis Dot structure for the valence shell of each element.
- Rearrange the electrons to pair up electrons from each atom.
- Draw circles to show the sharing of electrons between each pair of atoms.
- Draw the bond structure using chemical symbols and lines. Use one line for each pair of electrons that is shared.
- Write the chemical formula for each molecule.

a) Hydrogen + Hydrogen



b) Chlorine + Chlorine



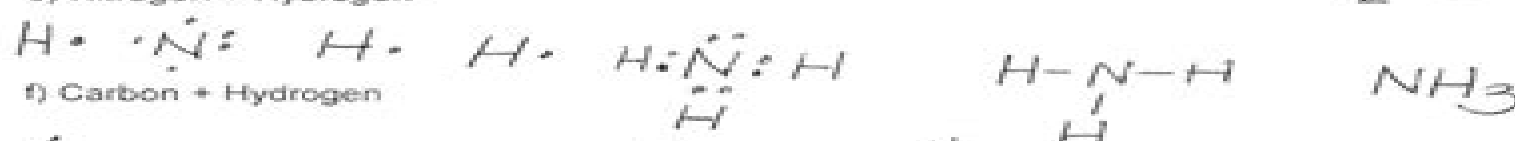
c) Hydrogen + Chlorine



d) Hydrogen + Oxygen



e) Nitrogen + Hydrogen



f) Carbon + Hydrogen



Covalent Bond Explore Learning Answer Key

**David Gosser, Victor S. Strozak, Mark S.
Cracolice**



Covalent Bond Explore Learning Answer Key:

Chemistry Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

Exploring Chemistry (Loose-Leaf) Matthew Johll, 2012-05-04 Matthew Johll s Exploring Chemistry covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It s the same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book s specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more EMRS PGT Chemistry Exam Book (English Edition) - Eklavya Model Residential School Post Graduate Teacher - 10 Practice Tests (1500 Solved Questions) EduGorilla Prep Experts, 2023-10-01 Best Selling Book in English Edition for EMRS PGT Post Graduate Teacher Chemistry Exam with objective type questions as per the latest syllabus Compare your performance with other students using Smart Answer Sheets in EduGorilla s EMRS PGT Post Graduate Teacher Chemistry Exam Practice Kit EMRS PGT Post Graduate Teacher Chemistry Exam Preparation Kit comes with 10 Practice Tests with the best quality content Increase your chances of selection by 16X EMRS PGT Post Graduate Teacher Chemistry Exam Prep Kit comes with well structured and 100% detailed solutions for all the questions Clear exam with good grades using thoroughly Researched Content by experts *Chemistry3* Andrew Burrows, John Holman, Andrew Parsons, Gwen Pilling, Gareth Price, 2017 Chemistry3 establishes the fundamental principles of all three strands of chemistry organic inorganic and physical By building on what students have learned at school using carefully worded explanations annotated diagrams and worked examples it presents an approachable introduction to chemistry and its relevance to everyday life Discover Biology Michael Lee Cain, Carol Kaesuk Yoon, Anu Singh-Cundy, 2009 Discover Biology helps students become biologically literate students to progress from science to scientific literacy Chemical Interactions, 2005

Investigating Chemistry Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are

the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package Alcamo's Fundamentals of Microbiology Jeffrey C. Pommerville,2012-01-20 Ideal for allied health and pre nursing students Alcamo s Fundamentals of Microbiology Body Systems Second Edition retains the engaging student friendly style and active learning approach for which award winning author and educator Jeffrey Pommerville is known Thoroughly revised and updated the Second Edition presents diseases complete with new content on recent discoveries in a manner that is directly applicable to students and organized by body system A captivating art program includes more than 150 newly added and revised figures and tables while new feature boxes Textbook Cases serve to better illuminate key concepts Pommerville s acclaimed learning design format enlightens and engages students right from the start and new chapter conclusions round out each chapter leaving readers with a clear understanding of key concepts **Principles of Modern Chemistry** David W. Oxtoby,H. Pat Gillis,Laurie J. Butler,2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom *Biology* Sylvia S. Mader,2000-07 **British Education Index** ,2004 **Peer-Led Team Learning** David Gosser,Victor S. Strozak,Mark S. Cracolice,2006 The Workshop Program is a collaboration of teachers learning specialists and students that develops and implements a Peer Led Team Learning model of teaching science Students who have done well in the course are selected and trained for leadership roles These peer leaders guide the work of teams of 6 8 students to solve carefully structured problems in weekly Workshop sessions *Discover Science: Teacher's annotated edition* ,1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today s technological society Fundamentals of Chemistry David E. Goldberg,2001 Current Index to Journals in Education ,1996 **The Software Encyclopedia 2000** Bowker Editorial Staff,2000-05 **Proceedings of the National Convention** National Science Teachers Association,National Science Teachers Association. A report of the proceedings [of the] national convention,1959 **Six Ideas That Shaped Physics: Unit Q - Particles Behaves Like Waves** Thomas A Moore,2003-01-09 SIX IDEAS THAT SHAPED PHYSICS is the 21st century s alternative to traditional encyclopedic textbooks Thomas Moore designed SIX IDEAS to teach students to apply basic physical principles to realistic situations to solve

realistic problems to resolve contradictions between their preconceptions and the laws of physics to organize the ideas of physics into an integrated hierarchy *Journal of Chemical Education* ,1924 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Peer-led Team Learning*
David K. Gosser, Victor S. Strozak, Mark S. Cracolice, 2001 Workbook developed from the Workshop Chemistry Project which explored developed and applied the concept of peer led team learning in problem solving workshops in introductory chemistry courses

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