Name: _		Class:		Date:	ID: A	
Cp phy	sics - Spring Fi	nal Review (second s	emes	ter topics)		
	e Choice the choice that best	completes the statement or	r ansmy	ers the question.		
	 a. Energy is r b. Kinetic energy c. The number 	 b. Kinetic energy is added to the particles of the substance. c. The number of atoms and molecules in a substance changes. 				
	What happens to a. It increases b. It decrease	·.	n ideal c. d.	gas when it is heated from 0°C to lt remains constant. It is impossible to determine.	4°C?	
	Which of the for a. elastic ener b. temperatur	gy	e.	netic energy of atoms and molecul potential energy thermal equilibrium	es?	
	As the tempera a. thermal eq b. thermal en	ailibeiann.	c.	volume tends to increase due to thermal expansion.		
	What is the ten steam at 1 atm a. 0°F b. 273 K		c.	quilibrium with another system m 0 K 100°C	ade up of water and	
	Energy transfer the following p a. mass b. specific be	roperties?	c.	odies in thermal contact when the density temperature	y differ in which of	
	Which of the for a. heat b. internal en	ollowing terms describes a ergy	transfi c. d.	er of energy? temperature kinetic energy		
	 The use of fibe through what p a. conduction radiation 	rocess?	er wali e. d.	s of a building is intended to mini convection vaporization	mize heat transfer	
	9. How is energy transferred as heat always directed? a. from an object at low temperature to an object at high temperature b. from an object at high temperature to an object at low temperature c. from an object at low kinetic energy to an object at high kinetic energy d. from an object with higher mass to an object of lower mass					

- 10. What three properties of a substance affect the amount of energy transferred as heat to or from the substance?
 - a. volume, temperature change, specific heat capacity
 - b. density, temperature change, specific heat capacity
 - c. mass, temperature change, specific heat capacity
 - d. mass, temperature change, latent heat

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JR Anderson

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