

Question 1

1(a)(iii) The three states of matter are solid, liquid and gas.

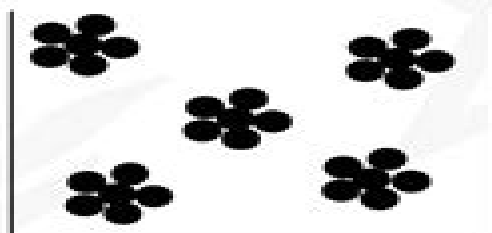
1(a)(iv) Melting point – the temperature at which a solid changes state to a liquid.

1(a)(v) The melting point of A is 65°C .

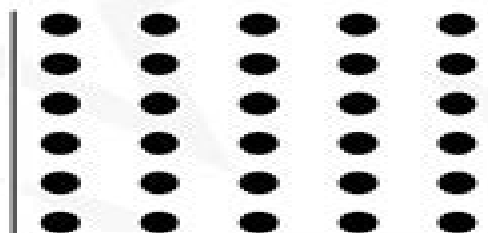
1(a)(vi) There would be a mixture of solid and liquid between $t = 2.8$ minutes and $t = 5.4$ minutes.

1(a)(vii) One precaution which should be taken during this experiment is that the boiling tube should be handled with a test tube holder, and not with bare/gloved hands.

1(a)(viii)



Particles of A at 0 min



Particles of A at 10 min

1(b)

	Inference
(i)	NO_2 gas produced; NO_3^- present.
(iii)	$\text{Ca}^{2+}_{(\text{aq})}$, $\text{Al}^{3+}_{(\text{aq})}$, $\text{Pb}^{2+}_{(\text{aq})}$ and/or $\text{Zn}^{2+}_{(\text{aq})}$ possibly present. $\text{Al}^{3+}_{(\text{aq})}$, $\text{Pb}^{2+}_{(\text{aq})}$ and/or $\text{Zn}^{2+}_{(\text{aq})}$ possibly present.
(iv)	$\text{Al}^{3+}_{(\text{aq})}$, $\text{Pb}^{2+}_{(\text{aq})}$ and/or $\text{Zn}^{2+}_{(\text{aq})}$ possibly present. $\text{Al}^{3+}_{(\text{aq})}$ or $\text{Pb}^{2+}_{(\text{aq})}$ possibly present.
(v)	$\text{Pb}^{2+}_{(\text{aq})}$ confirmed to be present. $\text{Pb}^{2+}_{(\text{aq})} + 2\text{I}^{-}_{(\text{aq})} \rightarrow \text{PbI}_{2(\text{s})}$.

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