

Name: _____ Date: _____ Class period: _____



Unit: Chemical Bonding **Answer Key** "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class. ☺

1. What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_2H_2

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: $H-H$

An electron dot formula uses dots instead of lines to show shared pairs of electrons.

Example: $H:H$

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a

Concepts For Today Answer Key

Horace William Brindley Joseph



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Issues for Today Lorraine C. Smith, Nancy Nici Mare, 1995 Issues for Today combines stimulating reading topics with a carefully planned combination of pre and post reading exercises and activities designed to reinforce comprehension and develop critical thinking skills High interest stimulating topics changing lifestyles crime dreams astronomy and language Thematically grouped readings three chapters per unit Additional opportunities for task based activities outside the classroom

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Critical

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The Vocational Guidance Magazine ,1926 The Personnel and Guidance Journal ,1926 The Concept of Method Gerhard Richard Lomer,1910 The Concept of Evolution Horace William Brindley Joseph,1924 Basic Concepts of Chemistry Leo J. Malone,Theodore O. Dolter,2012-04-13 Basic Concepts of Chemistry was originally written over thirty years ago to address the needs of general chemistry students with little or no background in chemistry Over time the text has evolved beyond purposes solely aimed at a preparatory chemistry course For some preparatory chemistry students a main sequence in general chemistry may follow but for others a semester of organic and biochemistry may follow Other students enroll to simply satisfy a basic science or chemistry requirement The text was written with a level and functionality designed to accommodate the needs of each of these varied groups of students Basic Concepts of Chemistry was designed with a flexibility that allows instructors to emphasize or omit certain clearly delineated sections The mission of the text has evolved in response to the increased diversity of students and the emphasis on outcomes assessment

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