

Chapter 32 Electrostatics

Exercises

32.1 Electrical Forces and Charges (pages 645-646)

1. Circle the letter beside the correct comparison of the strengths of the gravitational force and the electrical force.
 - a. The gravitational force is slightly stronger than the electrical force.
 - b. The electrical force is slightly stronger than the gravitational force.
 - c. The gravitational force is much stronger than the electrical force.
 - ☒ d. The electrical force is much stronger than the gravitational force.

2. Why don't you feel the electrical forces that act on you all the time?

So many around the cancel out

3. Describe the simple model of the atom proposed in the early 1900s by Rutherford and Bohr.

planetary

4. Charge is the fundamental electrical property to which the mutual attractions or repulsions between electrons or protons is attributed.

5. By convention, what is the charge of the following?

- a. electrons -
- b. protons +
- c. neutrons 0

6. Is the following sentence true or false? The mass of a proton is 2000 times greater than the mass of an electron. True

7. Circle the letter beside the correct comparison of the magnitudes of the charges of a proton and an electron.

- a. The magnitude of the proton's charge is slightly greater.
- b. The magnitude of the electron's charge is slightly greater.
- c. The magnitudes of a proton's charge and an electron's charge are always equal, but they vary for different atoms.
- ☒ d. The magnitudes of a proton's charge and an electron's charge are always equal and never change.

8. Like charges repel and opposite charges attract

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Raymond A. Serway



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Smart Cyber-Physical Power Systems, Volume 1 Ali Parizad, Hamid Reza Baghaee, Saifur Rahman, 2025-02-21 Authoritative highly comprehensive guide on how emerging technologies can address various challenges in different sectors of smart cyber physical power systems As the world shifts towards smarter and more resilient energy systems cyber physical power systems CPSs represent a critical step in modernizing the power infrastructure Smart Cyber Physical Power Systems Volume 1 Fundamental Concepts Challenges and Solutions offers an in depth exploration of the fundamental concepts structures and major challenges that underlie these complex systems It covers the essential theories and frameworks that drive the integration of digital technologies with physical power systems including smart grids microgrids and the Internet of Energy This volume addresses a range of crucial topics from global demand response strategies and microgrid architectures to smart energy management in cities and advanced distributed control strategies Additionally it highlights key challenges such as ensuring resiliency protecting against cyberattacks and maintaining reliability in the face of rapid technological advancements Experts from around the world contribute to this volume sharing vital insights into the transformation of traditional power systems into adaptive cyber physical networks Their focus on the growing importance of privacy security and data analytics makes this book a critical resource for anyone involved in power system research offering essential tools to navigate and shape the future landscapes of energy systems Whether you re a researcher engineer or industry professional this volume provides the foundational knowledge needed to understand the evolving landscape of smart cyber physical power systems and the significant challenges they face Join us on

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Computational Chemistry Errol G. Lewars, 2024-06-04 This is the fourth edition of the successful textbook on computational chemistry which continues to provide a comprehensive introduction to the theory and practice of computational chemistry Notable updates include a review of references up to mid 2023 encompassing recent developments in scientific journals books and software The evolving prominence of density functional theory DFT is emphasized and attention is given to the increasing application of artificial intelligence in computational chemistry The book maintains key features from the previous edition delving into the mathematical intricacies of ab initio and density functional methods at an introductory level Clear explanations of matrix methods are provided offering a direct approach to obtaining energy levels and molecular orbitals Additionally each chapter includes sets of Easier and Harder drill questions with suggested answers at the end of the book enhancing the learning experience The book is intended for upper year undergraduate and graduate students studying computational and theoretical chemistry and for self study by researchers in universities and industry to whom computational chemistry may be useful

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Conceptual Structures Timothy E. Nagle, 1992

Nuclear Science Abstracts, 1975

Essentials of Physics John D. Cutnell, Kenneth W. Johnson, 2006 No further information has been provided for this title

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