



Name: _____
Date: _____

COUNTING ATOMS

WORKSHEET

- Chemical formulas can have three components.
 - The _____ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
 - The _____ which are numbers that are found on the lower _____ - hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only _____ atom of this element in the molecule.
 - The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

4) Li ₂ SO ₄	
Element	# of Atoms
Total	

7) H ₂ COCH ₂	
Element	# of Atoms
Total	

10) NH ₄ C ₂ H ₃ O ₂	
Element	# of Atoms
Total	

13) Ca ₃ (PO ₄) ₂	
Element	# of Atoms
Total	

2) HNO ₃	
Element	# of Atoms
Total	

5) NH ₄ Cl	
Element	# of Atoms
Total	

8) Mg(OH) ₂	
Element	# of Atoms
Total	

11) NaC ₂ H ₃ O ₂	
Element	# of Atoms
Total	

14) Fe ₂ (CO ₃) ₃	
Element	# of Atoms
Total	

3) U ₂ O	
Element	# of Atoms
Total	

6) CaClO ₃	
Element	# of Atoms
Total	

9) Al(OH) ₃	
Element	# of Atoms
Total	

12) (NH ₄) ₃ PO ₄	
Element	# of Atoms
Total	

15) Al ₂ (SO ₄) ₃	
Element	# of Atoms
Total	

Counting Atoms Test

S Baum



Counting Atoms Test:

SAT Subject Test Chemistry Joseph A. Mascetta, Mark Kernion, 2020-12-01 The updated edition of Barron's SAT Subject Test Chemistry includes A full length diagnostic test with explained answers Four practice tests that reflect the actual SAT Subject Test Chemistry All questions answered and explained Detailed reviews covering all test topics Appendixes which include the Periodic Table important equation constant and data tables and a glossary of chemistry terms Both teachers and test taking students have praised earlier editions of this manual for its wealth of well organized detail Subject reviewed include the basics matter energy scientific method and measurements atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory ONLINE PRACTICE TESTS Students who purchase this book or package will also get access to two additional full length online SAT Chemistry subject tests with all questions answered and explained SAT Subject Test: Chemistry with Online Tests Joseph A. Mascetta, Mark Kernion, 2018-09-01 Always study with the most up to date prep Look for SAT Subject Test Chemistry ISBN 9781506263120 on sale December 01 2020 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entitles included with the product

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structure of physical science by emphasizing science as a search for truth rather than an accumulation of facts It develops the subject through concrete examples such as inclined planes and levers speeds and displacements progressing to consideration of forces and the concept of inertia and the idea of energy Similarly a study of observable chemical reactions advances to the ways in which atoms combine separate and displace one another and how observable masses of reactants and products illustrate how the atoms are combining The fundamental ideas are applied to astronomy optics geology music and the chemical compounds of life processes The wide variety of end of chapter problems and multiple choice questions reinforce comprehension of each topic

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Scientific American, 1913

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Officials,1986 *Journal of Chemical Education* ,1929 **Steam** ,1914 *Practical Chemistry* Lyman Churchill
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