

Name:

Representing the Bonding in Simple Covalent Molecules

Class

Covalent bonds are formed when non-metal atoms _____ of electrons between them, to achieve _____.

One shared pair is called a _____ bond, 2 shared pairs is called a _____ bond, and 3 shared pairs is called a _____ bond.

Name: Hydrogen

Formula: H_2

Dot Cross Diagram:



Displayed Formula:



Type of Covalent Bond:

Name: Fluorine

Formula:

Dot Cross Diagram:



Displayed Formula:

Type of Covalent Bond:

Name: Chlorine

Formula:

Dot Cross Diagram:



Displayed Formula:

Type of Covalent Bond:

Name: Bromine

Formula:

Dot Cross Diagram:



Displayed Formula:

Type of Covalent Bond:

Name: Hydrogen chloride

Formula:

Dot Cross Diagram:



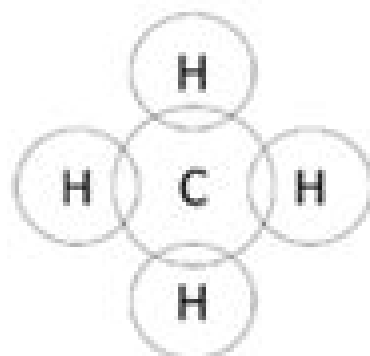
Displayed Formula:

Type of Covalent Bond:

Name: Methane

Formula:

Dot Cross Diagram:



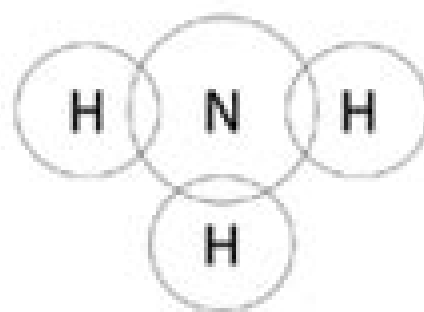
Displayed Formula:

Type of Covalent Bonds:

Name: Ammonia

Formula:

Dot Cross Diagram:



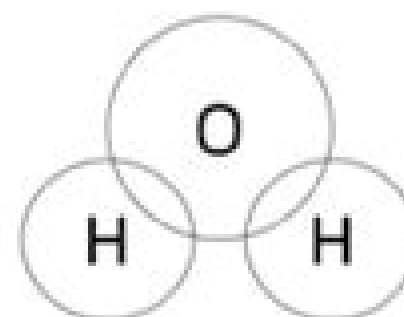
Displayed Formula:

Type of Covalent Bonds:

Name: Water

Formula:

Dot Cross Diagram:



Displayed Formula:

Type of Covalent Bonds:

Covalent Bonding Core Teaching Resources Test Key

Zhe Xu



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