Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

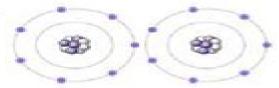
Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- There are eight markers in a full set, but Flora and Frank each only have seven markers.
 Flora is missing the red marker, and Frank is missing the blue marker. What can they do so
 that each has a full set of markers? Frank should give the red marker to Flora, and Flora
 should give the blue marker to Frank
- Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? Otto should give the black and brown markers to Olivia, and Olivia should give the purple and green markers to Otto

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo™, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



- The outermost electrons in each atom are called valence electrons. How many valence electrons does each fluorine atom have? 7 valence electrons
- Click Pause (). Drag an electron from the left atom to the right atom. Click Play.

What happens? They equally share the one atom on the new shell that is formed

Click Pause, drag an electron from the right atom to the left, and then click Play.

What happens now? They continue to spiral around both atoms, but now there are two electrons on the shell

Activity A: Sharing electrons	Get the Gizmo ready	
	Click Reset. Select Hydrogen.	(•)(

Covalent Bonding Exploration Guide Answers

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Teacher's Manual and Resource Guide for Exploring the Sciences Herbert Drapkin, 1964 **Computational** Approaches for Chemistry Under Extreme Conditions Nir Goldman, 2019-02-18 This book presents recently developed computational approaches for the study of reactive materials under extreme physical and thermodynamic conditions It delves into cutting edge developments in simulation methods for reactive materials including quantum calculations spanning nanometer length scales and picosecond timescales to reactive force fields coarse grained approaches and machine learning methods spanning microns and nanoseconds and beyond These methods are discussed in the context of a broad range of fields including prebiotic chemistry in impacting comets studies of planetary interiors high pressure synthesis of new compounds and detonations of energetic materials. The book presents a pedagogical approach for these state of the art approaches compiled into a single source for the first time Ultimately the volume aims to make valuable research tools accessible to experimentalists and theoreticians alike for any number of scientific efforts spanning many different types of Organic Chemistry, 5e Student Study Guide and Solutions Manual David R. compounds and reactive conditions Klein, Laurie S. Starkey, 2025-03-04 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry **Exploring Chemistry (Loose-Leaf)** Matthew Johll, 2012-05-04 Matthew Johll's Exploring Chemistry overs the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It s the same same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book s specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more Student Guide for Cycles of Life Gerarld L. Kellogg, 2006 Carolina Science and Math Carolina Biological Supply Company, 2003 **Chemistry Teacher's Guide** Michael P. Olmsted, 1972 **Fullerenes and Other** Carbon-Rich Nanostructures Jean-François Nierengarten, 2014-05-16 Yanfei Shen and Takashi Nakanishi Exotic Self Organized Fullerene Materials Based on Uncommon Hydrophobic Amphiphilic Approach Yuming Zhao and Guang Chen C60 Fullerene Amphiphiles as Supramolecular Building Blocks for Organized and Well Defined Nano scale Objects Anna Troeger Vito Sgobba and Dirk M Guldi Multilayer Assembly for Solar Energy Conversion Delphine Felder Flesch Self or Induced Organization of 60 Fullerene Hexakisadducts Andr s de la Escosura Olga Trukhina and Tom s Torres Dual Role of

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