#### You are to spend no more than ½ hour on this question.

(c)

A liquid, A, initially at temperature  $T_1 = 28.4$   $^{\circ}$ C, is heated by an immersion heater in a container. The resulting temperature,  $T_n$ , is recorded at 1 minute intervals and the following results obtained.

Time , t / min	0	1	2	3	4	5	-6	7
Temperature, T₂ / ° C	28.4	36.0	43.1	52.5	60.5	69.0	75.1	84.3
Temperature changes ΔT / °C								

(a)	Complete the table by	computing the	temperature changes	$\Delta T = T_i \cdot T_i$	(4 marks)
-----	-----------------------	---------------	---------------------	----------------------------	-----------

(b)	Plot a graph of	Temperature change,	ΔT, against time, t, on pag	pe 3. (9	marks)
-----	-----------------	---------------------	-----------------------------	----------	--------

(c)	Find the slope, S, of the graph.

(d)	$C_p$ the Specific Heat Capacity of the liquid is related to the slope S by: $C_p = \frac{2000}{s} \text{ J kg}^{-1} \text{ K}^{-1}$ . Find the Specific Heat Capacity of the liquid.	tarks)
		anneces:

What would the temperature of the liquid after 12 minutes?

(4 marks).

(2 marks)

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**Brian Ellis** 

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