

Density Laboratory

Answer Key

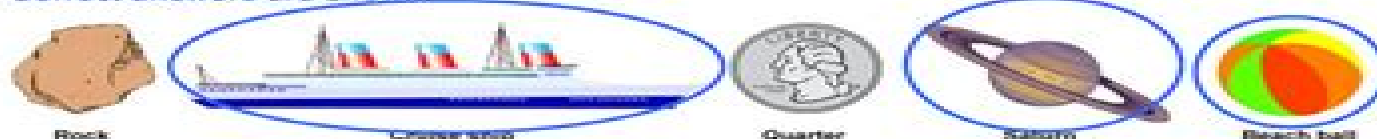
Vocabulary: buoyancy, density, graduated cylinder, mass, matter, scale, volume

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

[Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

1. Of the objects below, circle the ones you think would float in water. *Predictions will vary.*

Correct answers are shown:



2. Why do some objects float, while others sink?

Answers will vary. [An object floats when the weight of the liquid it displaces is equal to the weight of the object. Objects sink when they are heavier than the equivalent volume of liquid. In other words, objects sink when they are denser than the liquid.]

Gizmo Warm-up

The *Density Laboratory Gizmo™* allows you to measure a variety of objects, then drop them in water (or other liquid) to see if they sink or float.

1. An object's **mass** is the amount of **matter** it contains. The mass of an object can be measured with a calibrated **scale** like the one shown in the Gizmo. Drag the first object onto the **Scale**. (This is object 1.)

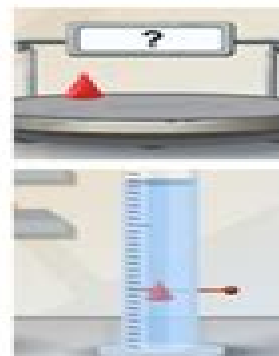
What is the mass of object 1? *19.5 grams (19.5 g)*

2. An object's **volume** is the amount of space it takes up. The volume of an irregular object can be measured by how much water it displaces in a **graduated cylinder**. Place object 1 into the **Graduated cylinder**.

What is the volume of object 1? *14.0 cm³ (equivalent to 14.0 mL)*

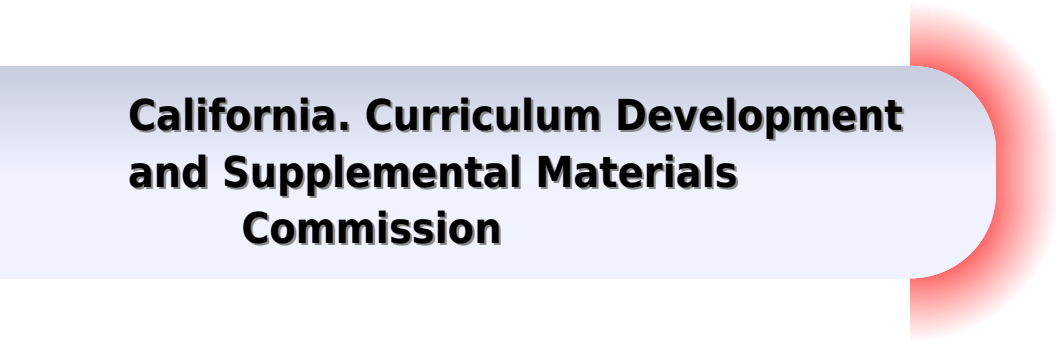
Note: While milliliters (mL) are used to measure liquid volumes, the equivalent unit cubic centimeters (cm³) are used for solids. Therefore, write the volume of object 1 in cm³.

3. Drag object 1 into the **Beaker of liquid**. Does it sink or float? *It sinks.*



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1. Understanding the eBook Density And Buoyancy Answer Keys
 - The Rise of Digital Reading Density And Buoyancy Answer Keys
 - Advantages of eBooks Over Traditional Books
2. Identifying Density And Buoyancy Answer Keys
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Density And Buoyancy Answer Keys
 - User-Friendly Interface
4. Exploring eBook Recommendations from Density And Buoyancy Answer Keys
 - Personalized Recommendations
 - Density And Buoyancy Answer Keys User Reviews and Ratings
 - Density And Buoyancy Answer Keys and Bestseller Lists
5. Accessing Density And Buoyancy Answer Keys Free and Paid eBooks
 - Density And Buoyancy Answer Keys Public Domain eBooks
 - Density And Buoyancy Answer Keys eBook Subscription Services
 - Density And Buoyancy Answer Keys Budget-Friendly Options
6. Navigating Density And Buoyancy Answer Keys eBook Formats

- ePub, PDF, MOBI, and More
- Density And Buoyancy Answer Keys Compatibility with Devices
- Density And Buoyancy Answer Keys Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Density And Buoyancy Answer Keys
 - Highlighting and Note-Taking Density And Buoyancy Answer Keys
 - Interactive Elements Density And Buoyancy Answer Keys
- 8. Staying Engaged with Density And Buoyancy Answer Keys
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Density And Buoyancy Answer Keys
- 9. Balancing eBooks and Physical Books Density And Buoyancy Answer Keys
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Density And Buoyancy Answer Keys
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Density And Buoyancy Answer Keys
 - Setting Reading Goals Density And Buoyancy Answer Keys
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Density And Buoyancy Answer Keys
 - Fact-Checking eBook Content of Density And Buoyancy Answer Keys
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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