General Chemistry II Hybrid Lab Course (CHEM 1108): Lab Report

CHEM 1108 - CRN: 11628 Date: 03/20/22

Student Name: Eterneti Ferguson, Romane Debon, Stuart

Davis

Experiment Number: 7

Lab Title: Weak acid-strong base titration

Objective: make molar solutions and titrate sodium hydroxide and acetic acid. Calculate

volume, temperature, and the ph.

Chemical Table:

Chemical Name	Chemical Formula	Molar Mass (g/mol)	Hazards	SDS Reference Number	PPE
Distilled water	H ₂ O	18 g/mol	Skin imitation	1	Lab coat, goggles
Sodium hydroxide	NaOH	39.997 g/mol	Severe burns to the eyes and skin	2	Lab coat, goggles, gloves
Acetic acid	CH,COOH	60.052 g/mol	Flammable liquid, severe burns, harmful if swallowed, skin irritation	3	Lab coat, goggles

References:

1: http://www.nano.pitt.edu/sites/default/files/MSDS/Gas/Distilled%20Water%20Gas.pdf

2: https://beta-

static.fishersci.com/content/dam/fishersci/en_US/documents/programs/education/regulator

y-documents/sds/chemicals/chemicals-s/S25881,pdf

3: https://beta-

static fisher sci. com/content/dam/fisher sci/en_US/documents/programs/education/regulator y-documents/sds/chemicals/chemicals-a/S25118.pdf

Equipment:

Type of Equipment (Include capacity wherever required)		Purpose
Volumetric flask	J	Use to make the 1 molar solutions.

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Marion Effie Potter

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Table of Contents Chem 1108 Lab Manual Answers

- 1. Understanding the eBook Chem 1108 Lab Manual Answers
 - The Rise of Digital Reading Chem 1108 Lab Manual Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chem 1108 Lab Manual Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem 1108 Lab Manual Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chem 1108 Lab Manual Answers
 - Personalized Recommendations
 - Chem 1108 Lab Manual Answers User Reviews and Ratings
 - Chem 1108 Lab Manual Answers and Bestseller Lists
- 5. Accessing Chem 1108 Lab Manual Answers Free and Paid eBooks
 - Chem 1108 Lab Manual Answers Public Domain eBooks
 - Chem 1108 Lab Manual Answers eBook Subscription Services
 - Chem 1108 Lab Manual Answers Budget-Friendly Options

- 6. Navigating Chem 1108 Lab Manual Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Chem 1108 Lab Manual Answers Compatibility with Devices
 - Chem 1108 Lab Manual Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chem 1108 Lab Manual Answers
 - Highlighting and Note-Taking Chem 1108 Lab Manual Answers
 - Interactive Elements Chem 1108 Lab Manual Answers
- 8. Staying Engaged with Chem 1108 Lab Manual Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem 1108 Lab Manual Answers
- 9. Balancing eBooks and Physical Books Chem 1108 Lab Manual Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem 1108 Lab Manual Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chem 1108 Lab Manual Answers
 - Setting Reading Goals Chem 1108 Lab Manual Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chem 1108 Lab Manual Answers
 - Fact-Checking eBook Content of Chem 1108 Lab Manual Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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