

CHEM LAB 117 Prelab quizzes experiments 1-10 FINAL

You measured a 10 mL sample of DI water using a buret and an analytical balance. The mass was determined to be 9.9979 g. What is the volume of the water if the temperature at the time of the experiment is 22°C? The density of water at 22°C is 0.997770 g/mL.

Enter the numerical value, without units, to five significant figures. - ✓✓10.0202

A clean dry beaker has an initial mass of 45.4730 g. You transfer 10.0107 mL of an unknown liquid to the beaker. The new mass of the beaker and liquid is 60.0125 g. What is the density of the unknown liquid?

Record your answer without units to four decimal places. - ✓✓1.4524

True or False: Density is a chemical property of matter. - ✓✓False

You are given an irregularly shaped object and determine that it has a mass of 38.4162 g. When placed in a graduated cylinder that initially contains 19.00 mL of water, the water level rises to 26.64 mL. What is the density of the unknown material?

Enter the numerical value, without units, to four significant figures. - ✓✓5.028

What is the density of an object, in g/cm³, if its volume is 53.96 cm³ and its mass is 102.33 g? Enter the numerical value without units. Your answer should be within 1% of the correct value. If you carry out calculations to 3 significant figure precision, your answer should be within this range. - ✓✓1.90

What is the volume of a cylinder, in cm³, if the cylinder's diameter is 3.2 cm and its height is 8.4 cm? Enter the numerical value, without units, to three significant figures. Your answer should be

Chem 117 Lab Manual Answers Experiment 1

Jason Evans



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