

Balancing Chemical Equations Worksheet

1. 2 H_2 + 1 $\text{O}_2 \rightarrow$ 2 H_2O
2. 1 N_2 + 3 $\text{H}_2 \rightarrow$ 2 NH_3
3. 1 S_8 + 12 $\text{O}_2 \rightarrow$ 8 SO_3
4. 2 N_2 + 1 $\text{O}_2 \rightarrow$ 2 N_2O
5. 2 $\text{HgO} \rightarrow$ 2 Hg + 1 O_2
6. 6 CO_2 + 6 $\text{H}_2\text{O} \rightarrow$ 1 $\text{C}_6\text{H}_{12}\text{O}_6$ + 6 O_2
7. 1 Zn + 2 $\text{HCl} \rightarrow$ 1 ZnCl_2 + 1 H_2
8. 1 SiCl_4 + 4 $\text{H}_2\text{O} \rightarrow$ 1 H_4SiO_4 + 4 HCl
9. 2 Na + 2 $\text{H}_2\text{O} \rightarrow$ 2 NaOH + 1 H_2
10. 2 $\text{H}_3\text{PO}_4 \rightarrow$ 1 $\text{H}_4\text{P}_2\text{O}_7$ + 1 H_2O
11. 1 $\text{C}_{10}\text{H}_{16}$ + 8 $\text{Cl}_2 \rightarrow$ 10 C + 16 HCl
12. 1 CO_2 + 2 $\text{NH}_3 \rightarrow$ 1 $\text{OC}(\text{NH}_2)_2$ + 1 H_2O
13. 4 Si_2H_3 + 17 $\text{O}_2 \rightarrow$ 8 SiO_2 + 6 H_2O_3
14. 2 $\text{Al}(\text{OH})_3$ + 3 $\text{H}_2\text{SO}_4 \rightarrow$ 1 $\text{Al}_2(\text{SO}_4)_3$ + 6 H_2O
15. 4 Fe + 3 $\text{O}_2 \rightarrow$ 2 Fe_2O_3
16. 1 $\text{Fe}_2(\text{SO}_4)_3$ + 6 $\text{KOH} \rightarrow$ 3 K_2SO_4 + 2 $\text{Fe}(\text{OH})_3$
17. 2 $\text{C}_7\text{H}_6\text{O}_2$ + 15 $\text{O}_2 \rightarrow$ 14 CO_2 + 6 H_2O
18. 1 H_2SO_4 + 8 $\text{HI} \rightarrow$ 1 H_2S + 4 I_2 + 4 H_2O
19. 4 FeS_2 + 11 $\text{O}_2 \rightarrow$ 2 Fe_2O_3 + 8 SO_2
20. 2 Al + 3 $\text{FeO} \rightarrow$ 1 Al_2O_3 + 3 Fe
21. 1 Fe_2O_3 + 3 $\text{H}_2 \rightarrow$ 2 Fe + 3 H_2O
22. 1 Na_2CO_3 + 2 $\text{HCl} \rightarrow$ 2 NaCl + 1 H_2O + 1 CO_2
23. 2 K + 1 $\text{Br}_2 \rightarrow$ 2 KBr
24. 1 C_7H_{16} + 11 $\text{O}_2 \rightarrow$ 7 CO_2 + 8 H_2O
25. 1 P_4 + 5 $\text{O}_2 \rightarrow$ 2 P_2O_5

Chem 130 Balancing Equations Answers

David M. Himmelblau, James B. Riggs



Chem 130 Balancing Equations Answers:

Calculations for A-level Chemistry E. N. Ramsden, 2001 Aiming to match the various specifications this book gives explanations worked examples and practice in chemistry calculations It includes a comprehensive mathematics foundation section Work on formulae and equations the mole volumetric analysis and other key areas are included It is useful as a course book as well as for exam practice Access to Chemistry Avril Higton, Mike Clemmet, Elaine Golding, Alan V Jones, 2007-10-31 This text has been specifically designed to prepare people with previously limited chemical knowledge for entrance into science related courses such as Foundation and Access courses which involve chemistry in higher education Until now there have been no texts available for use on these courses and this book fills that gap Access to Chemistry effectively forms a self study course which is split into separate modules and units covering the full spread of concepts required for those needing a basic knowledge of chemistry The material is presented in a friendly and easy to use manner which allows the student to pace their acquisition of knowledge and gain increasing confidence in order to succeed in understanding essential relevant concepts Other useful features of this book include starter diagnostic tests worked examples and self study tests with answers at the end of each unit In addition to Access or Foundation course students and their tutors to whom this book will prove essential it will have an appeal also as a revision text for those needing a refresher after a break in the subject In addition it will be of interest to members of the general public who wish to better educate themselves on chemical matters as it provides a clear and useful insight into areas such as health home chemicals business market trends and gardening **Syllabus for Chemistry 130-230** University of Chicago, 1935 **Chemical Problem-solving by Dimensional Analysis** Arnold B. Loebel, 1978 **2015 / 2016 ASVAB For Dummies with Online Practice** Rod Powers, 2015-04-20 Ace the ASVAB with this comprehensive easy to use guide Expanded and improved 2015 2016 ASVAB For Dummies with Online Practice offers everything you ll need to hone your skills before taking the big test In this easy to use complete guide expert Rod Powers a retired First Sergeant from the U S Air Force covers the Armed Services Vocational Aptitude Battery like no one else can The book offers in depth reviews of all nine test subjects along with complete explanations for every question To give you the opportunity to sharpen your test taking skills the guide also offers exercises and tips that will help you pinpoint the areas where you need the most help To make sure you re as prepared as can be access to practice tools online takes things one step further featuring over one thousand questions that allow you to create your own optimized exams so you ll be ready on test day Offers comprehensive preparation with six full length ASVAB practice tests that will sharpen your test taking skills Provides two AFQT practice tests to assess enlistment eligibility Gives you the ability to create your own tests for practice where you really need it along with feedback on your progress Offers 300 flashcards to improve your vocabulary Jump in to 2015 2016 ASVAB For Dummies and take your military test taking skills to the next level *Revise AS Chemistry for Salters (OCR)* Ann Daniels, 2005-02-22 Help students pull together the chemical

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Mathematical Preparation for General Chemistry William L. Masterton,Emil J. Slowinski,1982

Basic Principles and Calculations in Chemical Engineering David M. Himmelblau,James B. Riggs,2012-05-31 The Number One Guide to Chemical Engineering Principles Techniques Calculations and Applications Now Even More Current Efficient and Practical Basic Principles and Calculations in Chemical Engineering Eighth Edition goes far beyond traditional introductory chemical engineering topics presenting applications that reflect the full scope of contemporary chemical petroleum and environmental engineering Celebrating its fiftieth Anniversary as the field s leading practical introduction it has been extensively updated and reorganized to cover today s principles and calculations more efficiently and to present far more coverage of bioengineering nanoengineering and green engineering Offering a strong foundation of skills and knowledge for successful study and practice it guides students through formulating and solving material and energy balance problems as well as describing gases liquids and vapors Throughout the authors introduce efficient consistent student friendly methods for solving problems analyzing data and gaining a conceptual application based understanding of modern chemical engineering processes This edition s improvements include many new problems examples and homework assignments Coverage includes Modular chapters designed to support introductory chemical engineering courses of any length Thorough introductions to unit conversions basis selection and process measurements Consistent sound strategies for solving material and energy balance problems Clear introductions to key concepts ranging from stoichiometry to enthalpy Behavior of gases liquids and solids ideal real gases single component two

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