

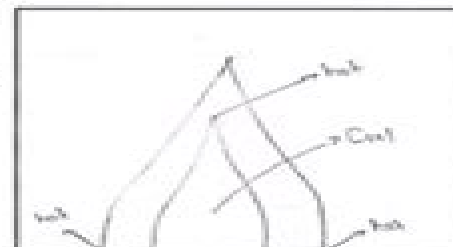
Experiment 1 Report Sheet

Basic Laboratory Operations

A. Bunsen Burner

1. Instructor's Approval of a well-adjusted Bunsen flame: _____
2. a. At right, draw a sketch of the heat zones for a nonluminous flame as directed with the wire gauze positioned parallel to the burner barrel. Label the "cool" and "hot" zones.
 - b. What happens to the "cool" and "hot" zones of a nonluminous flame, when the air-control valve is closed (nearly closed)? Explain.

As the air-control valve approaches the nearly closed position, the "cool" and "hot" zones merge into one flame.



B. Laboratory Balances

1. Determine the mass of the following objects on your assigned balance. Express your results with the correct sensitivity.

Balance no. _____ Sensitivity 0.001 g

Object	Description of Object (size or volume)	Mass (g)
Test tube		<u>5.240 g</u>
Beaker	<u>50 mL</u>	<u>29.052 g</u>
Spatula		<u>7.401 g</u>
Graduated cylinder	<u>10 mL</u>	<u>15.711 g</u>
Other (see instructor)	<u>Unknown #2</u>	<u>2.135 g</u>

2. Precision of Measurement. Balance no. _____

Mass of graduated cylinder (if not tared) 15.711 g

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5	Trial 6
a. Mass of cylinder + H ₂ O (l) (g)	<u>20.261 g</u>	<u>25.423 g</u>	<u>21.002 g</u>			
b. Mass of 7 mL H ₂ O (l) (g)	<u>4.55 g</u>	<u>4.71 g</u>	<u>4.511 g</u>			
c. Average mass of 7 mL H ₂ O (l) (g)			<u>4.59 g</u>			

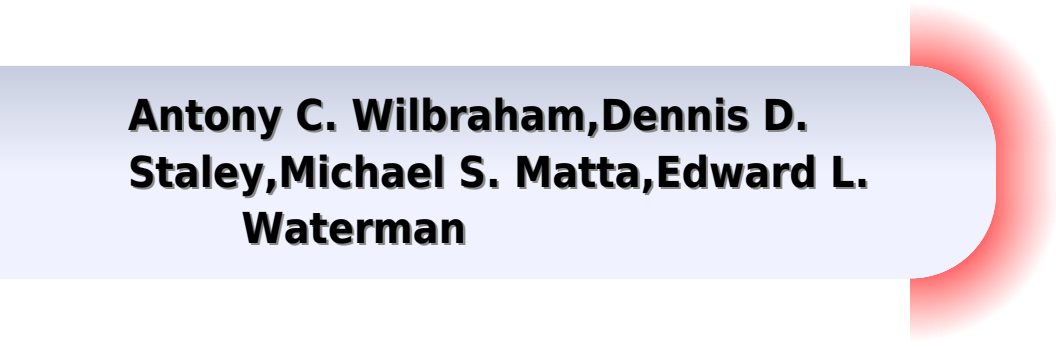
Comment on the precision of the mass measurements for the water samples.

Trial #3 does not seem to be precise.

Date Analysis, B

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**Antony C. Wilbraham, Dennis D.
Staley, Michael S. Matta, Edward L.
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 - The Rise of Digital Reading Chem 1blab Manual Answers Fresno State
 - Advantages of eBooks Over Traditional Books
2. Identifying Chem 1blab Manual Answers Fresno State
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem 1blab Manual Answers Fresno State
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chem 1blab Manual Answers Fresno State
 - Personalized Recommendations
 - Chem 1blab Manual Answers Fresno State User Reviews and Ratings
 - Chem 1blab Manual Answers Fresno State and Bestseller Lists
5. Accessing Chem 1blab Manual Answers Fresno State Free and Paid eBooks

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 - Chem 1blab Manual Answers Fresno State Budget-Friendly Options
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 - Chem 1blab Manual Answers Fresno State Enhanced eBook Features
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- Adjustable Fonts and Text Sizes of Chem 1blab Manual Answers Fresno State
 - Highlighting and Note-Taking Chem 1blab Manual Answers Fresno State
 - Interactive Elements Chem 1blab Manual Answers Fresno State
8. Staying Engaged with Chem 1blab Manual Answers Fresno State
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem 1blab Manual Answers Fresno State
9. Balancing eBooks and Physical Books Chem 1blab Manual Answers Fresno State
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem 1blab Manual Answers Fresno State
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chem 1blab Manual Answers Fresno State
- Setting Reading Goals Chem 1blab Manual Answers Fresno State
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chem 1blab Manual Answers Fresno State
- Fact-Checking eBook Content of Chem 1blab Manual Answers Fresno State
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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