

Qualitative Analysis and Chemical Bonding

Concepts:

- Chemical bonding
- Covalent bonding
- Ionic bonding
- Metallic bonding
- Intermolecular bonding
- Physical & chemical properties
- Electronegativity

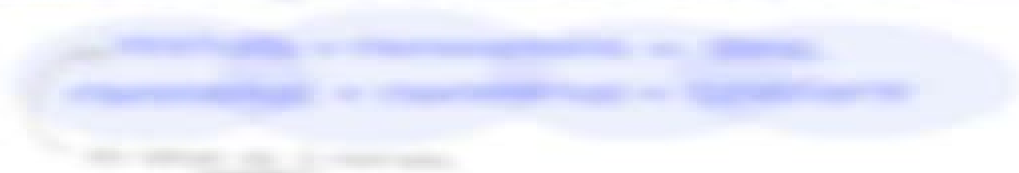
Introduction

- Types of atoms are not equal in chemical bond: **electronegativity**
- In order to understand bond, all atoms/molecules are equal in this. There are two types of chemical bonding.
- In order to understand chemical bonding, there are two types of atoms in nature. In this case, we will discuss the types of atoms in nature.

- Electronegativity:** tendency of an atom to attract electrons towards itself. It is a measure of the ability of an atom to attract electrons towards itself. It is a measure of the ability of an atom to attract electrons towards itself.
- Electronegativity is a measure of the ability of an atom to attract electrons towards itself. It is a measure of the ability of an atom to attract electrons towards itself.
- Electronegativity is a measure of the ability of an atom to attract electrons towards itself. It is a measure of the ability of an atom to attract electrons towards itself.



- Chemical bonding:** the process of atoms combining their outer shell of electrons to form a stable chemical bond.
- Electronegativity:** a measure of the ability of an atom to attract electrons towards itself.



Electronegativity is a measure of the ability of an atom to attract electrons towards itself. It is a measure of the ability of an atom to attract electrons towards itself.

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