

- 2) $\text{___ K} + \text{___ B}_2\text{O}_3 \rightarrow \text{___ K}_2\text{O} + \text{___ B}$
- 3) $\text{___ HCl} + \text{___ NaOH} \rightarrow \text{___ NaCl} + \text{___ H}_2\text{O}$
- 4) $\text{___ Na} + \text{___ NaNO}_3 \rightarrow \text{___ Na}_2\text{O} + \text{___ N}_2$
- 5) $\text{___ C} + \text{___ S}_8 \rightarrow \text{___ CS}_2$
- 6) $\text{___ Na} + \text{___ O}_2 \rightarrow \text{___ Na}_2\text{O}$
- 7) $\text{___ N}_2 + \text{___ O}_2 \rightarrow \text{___ N}_2\text{O}_5$
- 8) $\text{___ H}_3\text{PO}_4 + \text{___ Mg(OH)}_2 \rightarrow \text{___ Mg}_3(\text{PO}_4)_2 + \text{___ H}_2\text{O}$
- 9) $\text{___ NaOH} + \text{___ H}_2\text{CO}_3 \rightarrow \text{___ Na}_2\text{CO}_3 + \text{___ H}_2\text{O}$
- 10) $\text{___ KOH} + \text{___ HBr} \rightarrow \text{___ KBr} + \text{___ H}_2\text{O}$
- 11) $\text{___ Na} + \text{___ O}_2 \rightarrow \text{___ Na}_2\text{O}$
- 12) $\text{___ Al(OH)}_3 + \text{___ H}_2\text{CO}_3 \rightarrow \text{___ Al}_2(\text{CO}_3)_3 + \text{___ H}_2\text{O}$
- 13) $\text{___ Al} + \text{___ S}_8 \rightarrow \text{___ Al}_2\text{S}_3$
- 14) $\text{___ Cs} + \text{___ N}_2 \rightarrow \text{___ Cs}_3\text{N}$
- 15) $\text{___ Mg} + \text{___ Cl}_2 \rightarrow \text{___ MgCl}_2$
- 16) $\text{___ Rb} + \text{___ RbNO}_3 \rightarrow \text{___ Rb}_2\text{O} + \text{___ N}_2$
- 17) $\text{___ C}_6\text{H}_6 + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
- 18) $\text{___ N}_2 + \text{___ H}_2 \rightarrow \text{___ NH}_3$
- 19) $\text{___ C}_{10}\text{H}_{22} + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
- 20) $\text{___ Al(OH)}_3 + \text{___ HBr} \rightarrow \text{___ AlBr}_3 + \text{___ H}_2\text{O}$
- 21) $\text{___ CH}_3\text{CH}_2\text{CH}_2\text{CH}_3 + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
- 22) $\text{___ C}_2\text{H}_6 + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
- 23) $\text{___ Li} + \text{___ AlCl}_3 \rightarrow \text{___ LiCl} + \text{___ Al}$
- 24) $\text{___ C}_2\text{H}_6 + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
- 25) $\text{___ NH}_4\text{OH} + \text{___ H}_3\text{PO}_4 \rightarrow \text{___ (NH}_4)_3\text{PO}_4 + \text{___ H}_2\text{O}$
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- 29) $\text{___ Na} + \text{___ Cl}_2 \rightarrow \text{___ NaCl}$
- 30) $\text{___ Rb} + \text{___ S}_8 \rightarrow \text{___ Rb}_2\text{S}$
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- 32) $\text{___ NH}_3 + \text{___ HCl} \rightarrow \text{___ NH}_4\text{Cl}$
- 33) $\text{___ Li} + \text{___ H}_2\text{O} \rightarrow \text{___ LiOH} + \text{___ H}_2$
- 34) $\text{___ Ca}_3(\text{PO}_4)_2 + \text{___ SiO}_2 + \text{___ C} \rightarrow \text{___ CaSiO}_3 + \text{___ CO} + \text{___ P}$
- 35) $\text{___ NH}_3 + \text{___ O}_2 \rightarrow \text{___ N}_2 + \text{___ H}_2\text{O}$

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Mazhar Iqbal



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