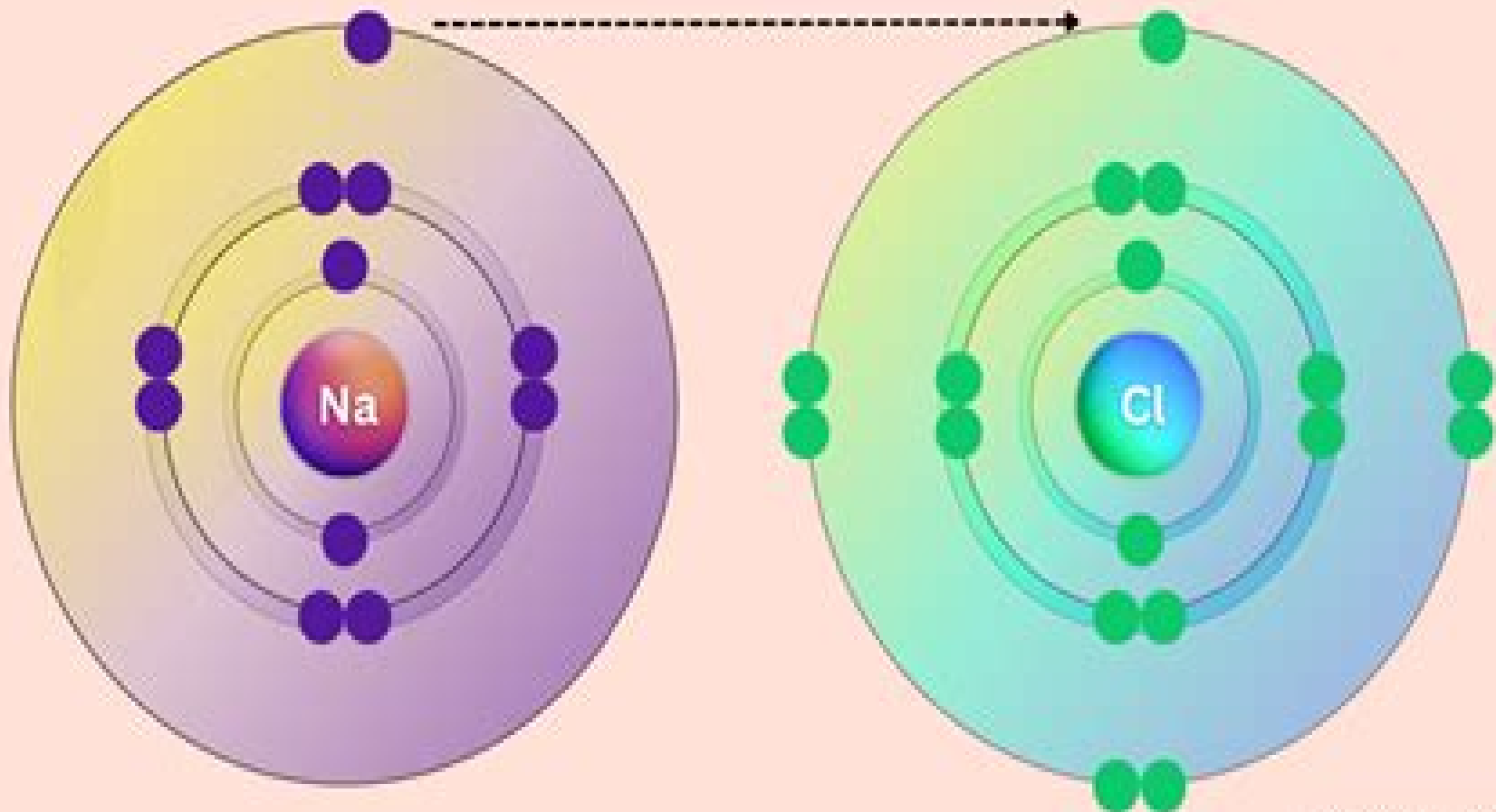


Ionic Bond Definition and Example

An ionic bond is an electrostatic attraction between two atoms where one atom transfers an electron to the other atom.



Chemactive Ionic Bonding

Nathaniel Warren Alcock

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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and Crystal Structures **Ionic Bonds and Compounds** Roberts,1997-02-01 *Structure and Bonding* Jack Barrett,2001
This book explains in non mathematical terms where possible the factors that govern covalent bond formation the lengths and strengths of bonds and molecular shapes **The Chemical Bond** Gernot Frenking,Sason Shaik,2014-06-13 A unique overview of the different kinds of chemical bonds that can be found in the periodic table from the main group elements to transition elements lanthanides and actinides It takes into account the many developments that have taken place in the field over the past few decades due to the rapid advances in quantum chemical models and faster computers This is the perfect complement to Chemical Bonding Fundamentals and Models by the same editors who are two of the top scientists working on this topic each with extensive experience and important connections within the community The Chemical Bond II D. Michael P. Mingos,2016-06-18 The series Structure and Bonding publishes critical reviews on topics of research concerned with chemical structure and bonding The scope of the series spans the entire Periodic Table and addresses structure and bonding issues associated with all of the elements It also focuses attention on new and developing areas of modern structural and theoretical chemistry such as nanostructures molecular electronics designed molecular solids surfaces metal clusters and supramolecular structures Physical and spectroscopic techniques used to determine examine and model structures fall within the purview of Structure and Bonding to the extent that the focus is on the scientific results obtained and not on specialist information concerning the techniques themselves Issues associated with the development of bonding models and generalizations that illuminate the reactivity pathways and rates of chemical processes are also relevant The individual volumes in the series are thematic The goal of each volume is to give the reader whether at a university or in industry a comprehensive overview of an area where new insights are emerging that are of interest to a larger scientific audience Thus each review within the volume critically surveys one aspect of that topic and places it within the context of the volume as a whole The most significant developments of the last 5 to 10 years should be presented using selected examples to illustrate the principles discussed A description of the physical basis of the experimental techniques that have been used to provide the primary data may also be appropriate if it has not been covered in detail elsewhere The coverage need not be exhaustive in data but should rather be conceptual concentrating on the new principles being developed that will allow the reader who is not a specialist in the area covered to understand the data presented Discussion of possible future research directions in the area is welcomed Review articles for the individual volumes are invited by the volume editors The Chemical Bond in Inorganic Chemistry I David Brown,2016-09-29 The bond valence model a description of acid base bonding is widely used for analysing and modelling the structures and properties of solids and liquids Unlike other models of inorganic chemical bonding the bond valence model is simple intuitive and predictive and is accessible to anyone with a pocket calculator and a secondary school command of chemistry and physics This new edition of The Chemical Bond in Inorganic Chemistry The Bond Valence Model shows how chemical properties arise naturally from the conflict between the constraints of chemistry

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Chemistry of Chemical Bonding R. K. Sharma, 2007

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